

High Temperature Oxidation Issues in Fossil Boilers

B. A. Pint, M. A. Bestor, S. Dryepondt
Oak Ridge National Laboratory
Oak Ridge, TN 37831-6156

Y. Zhang
Tennessee Technological University
Cookeville, TN 38505

Research sponsored by the U.S. Dept. of Energy
Fossil Energy Advanced Research Materials Program

Acknowledgments

ORNL

A. Haynes, K. Cooley - CVD coating fabrication

K. Unocic - characterization

G. Garner, J. Moser, T. Brummett, M. Stephens, M. Howell -
oxidation experiments

H. Longmire, K. Thomas - metallography

T. Brummett - SEM

L. Walker - EPMA

Tennessee Tech.

B. Bates - SEM

Research sponsored by the U. S. Department of Energy,
Fossil Energy Advanced Materials Research Program

Outline/Introduction

FY00-FY09: Al-rich (aluminide) coating study
+ Fe-Al alloy optimization study

Oxidation-resistant coating background:

Al-rich (aluminide) coatings very promising

Advanced ultra-supercritical steam - up to 760°C

Coal gasification - low P_{O_2} , high P_{S_2}

In the US, Al-rich coatings **not used** in boilers

Extensive research by Rapp, ORNL, etc.

Field tests in 80's **unsuccessful**

ORNL objective - **define benefits/address issues**

model lifetime (min 40kh), determine max. T

FY10 **Effect of oxy-firing on corrosion**
mechanisms in coal-fired boilers

Coatings: Last 10 years at ORNL

Fabrication - chemical vapor deposition (CVD)

- not for boilers or commercialization
- make clean, uniform coatings for research
- full control of process, no “black box”

Substrates - representative Fe-base alloys

T91 - Fe-9Cr-1Mo (later T92, T122)

304L - low C Fe-18Cr-8Ni (later 316)

Diffusion - define substrate interaction

exposures at 500°-800°C for 2-10kh

Oxidation - obvious benefit

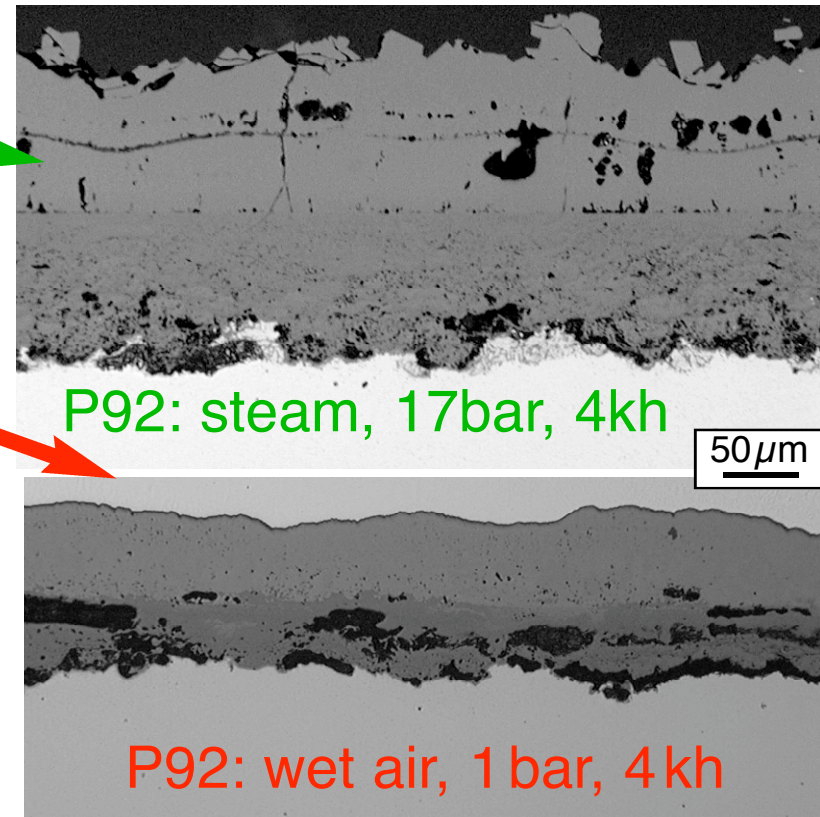
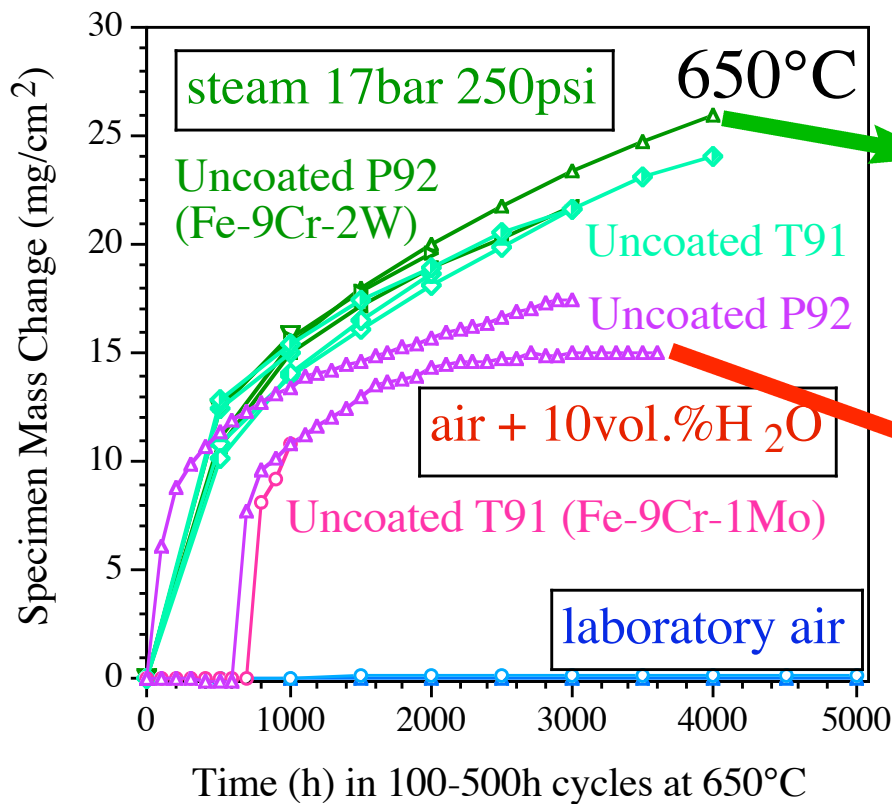
define failure criteria (critical Al at failure: C_b)

accelerated testing at 700°-800°C

testing: wet air, recently in 17bar steam

Fe-9Cr in Steam vs. Humid Air

comparison of mass gain and reaction products
650°C, 1202°F



Similar attack in steam and wet air (10±1 vol.% H₂O)

Define failure: must have environment that attacks substrate

Prior work in lab. air could not define coating lifetime

Key Points

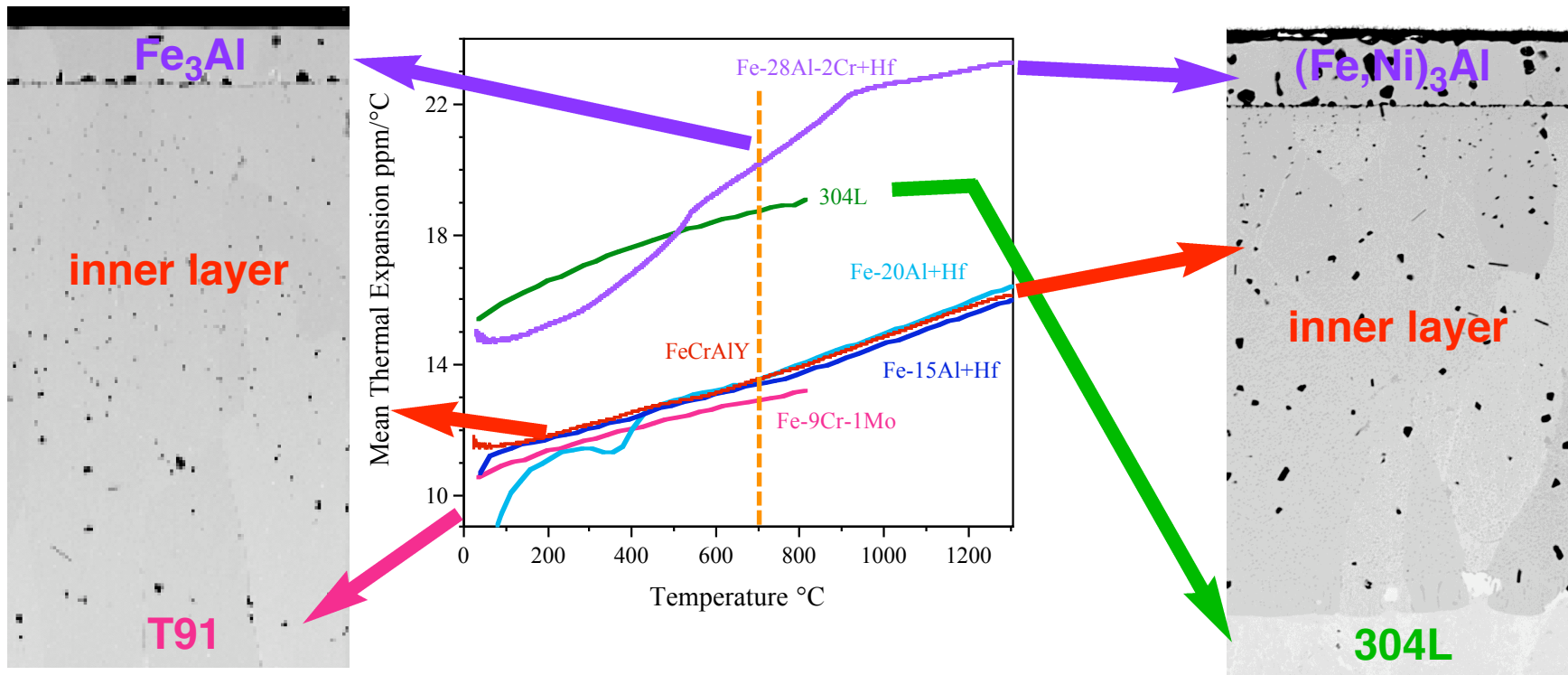
Normally think that more AI is better
thermal expansion mismatch problem

Intrinsic Aluminide Coating Problem

Substrate-coating thermal expansion mismatch

Ferritic

Austenitic



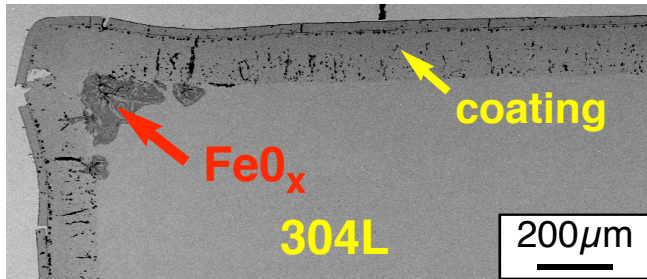
For coated **304L**, 3 layers with 3 different CTE

For coated **T91**, 3 layers with thick (250 μ m) coat

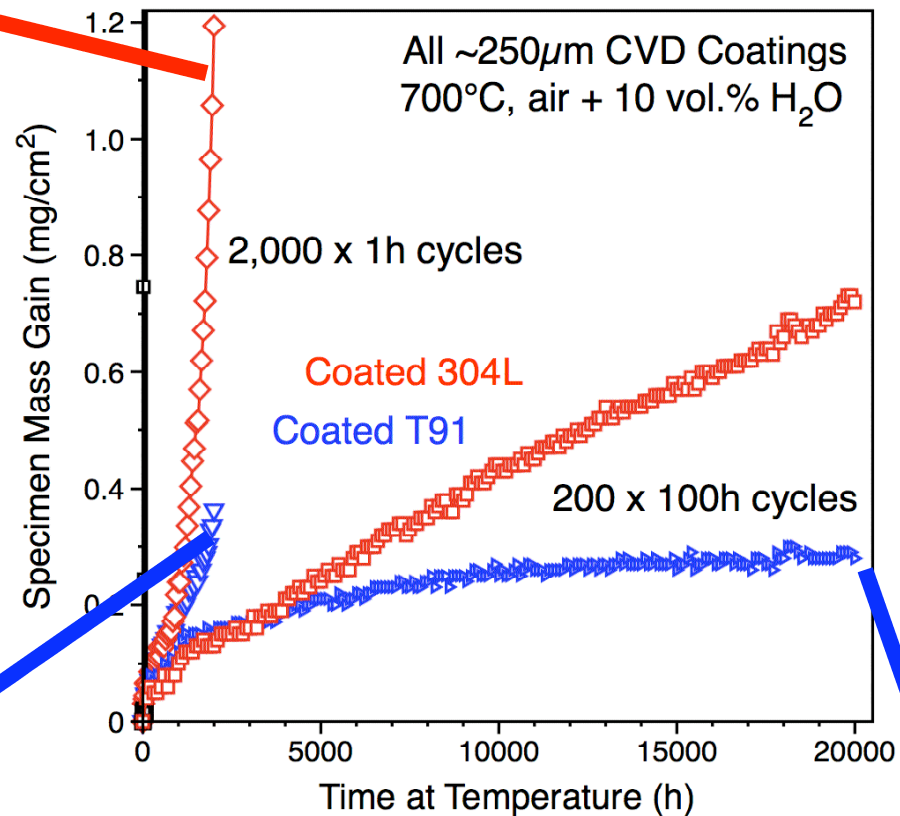
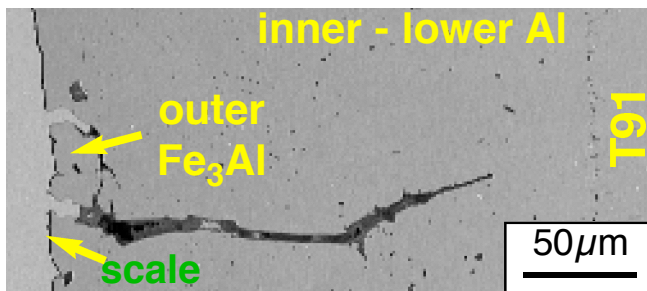
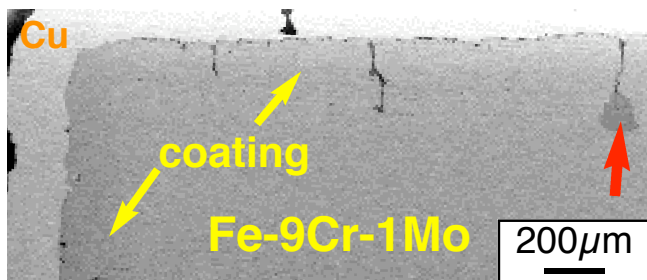
BUT, “thin” coating (50 μ m) no Fe₃Al - small Δ CTE

If CTE mismatch is problem, can thermal cycling crack coating? **YES**

Thick CVD coatings, 1h & 100h cycles, 700°C, humid air



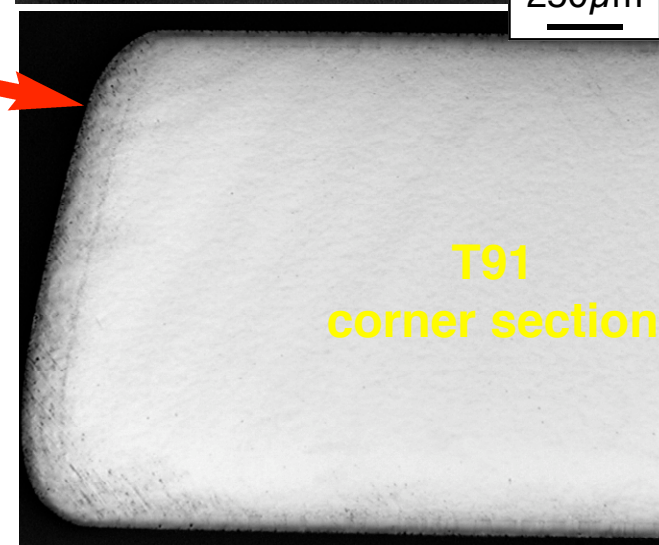
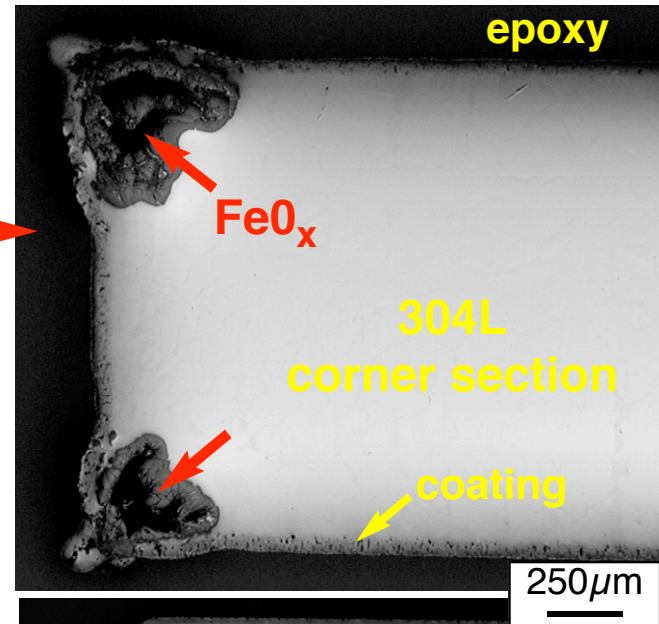
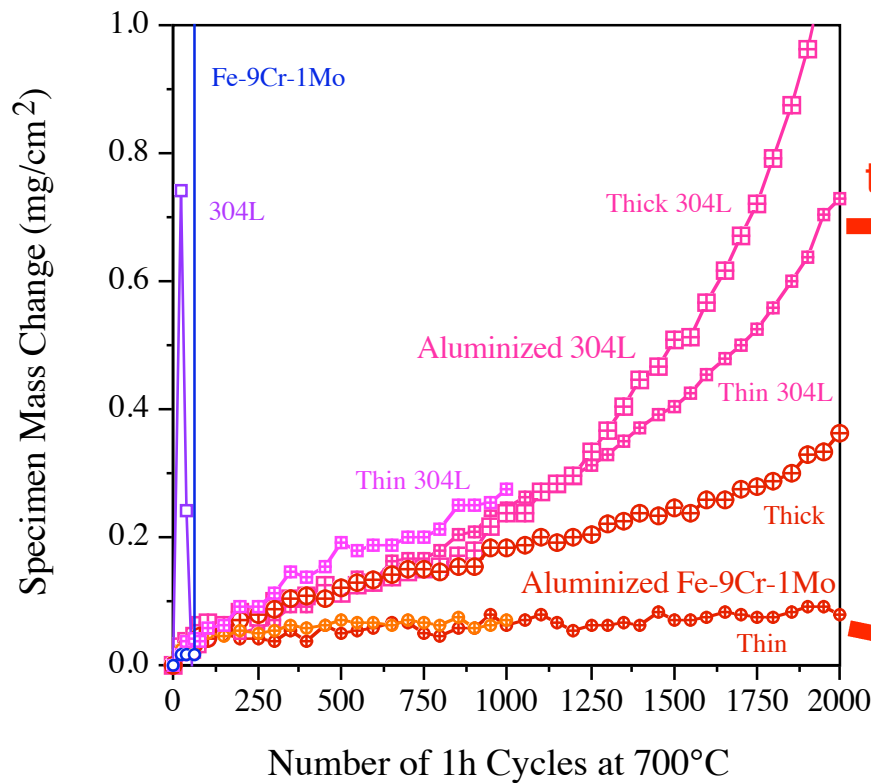
Coating cracking = substrate attack



100h cycles stopped at 20kh: no attack

If CTE mismatch is problem, does thickness (δ) affect performance? **Yes**

Thin & thick CVD coatings, **1h cycles**, 700°C, humid air



$$\text{stress} = f(\Delta\alpha_{\text{coat-sub}})$$

$$W = f(\delta_{\text{coating}})$$

(strain energy)

Key Points

Normally think that more Al is better

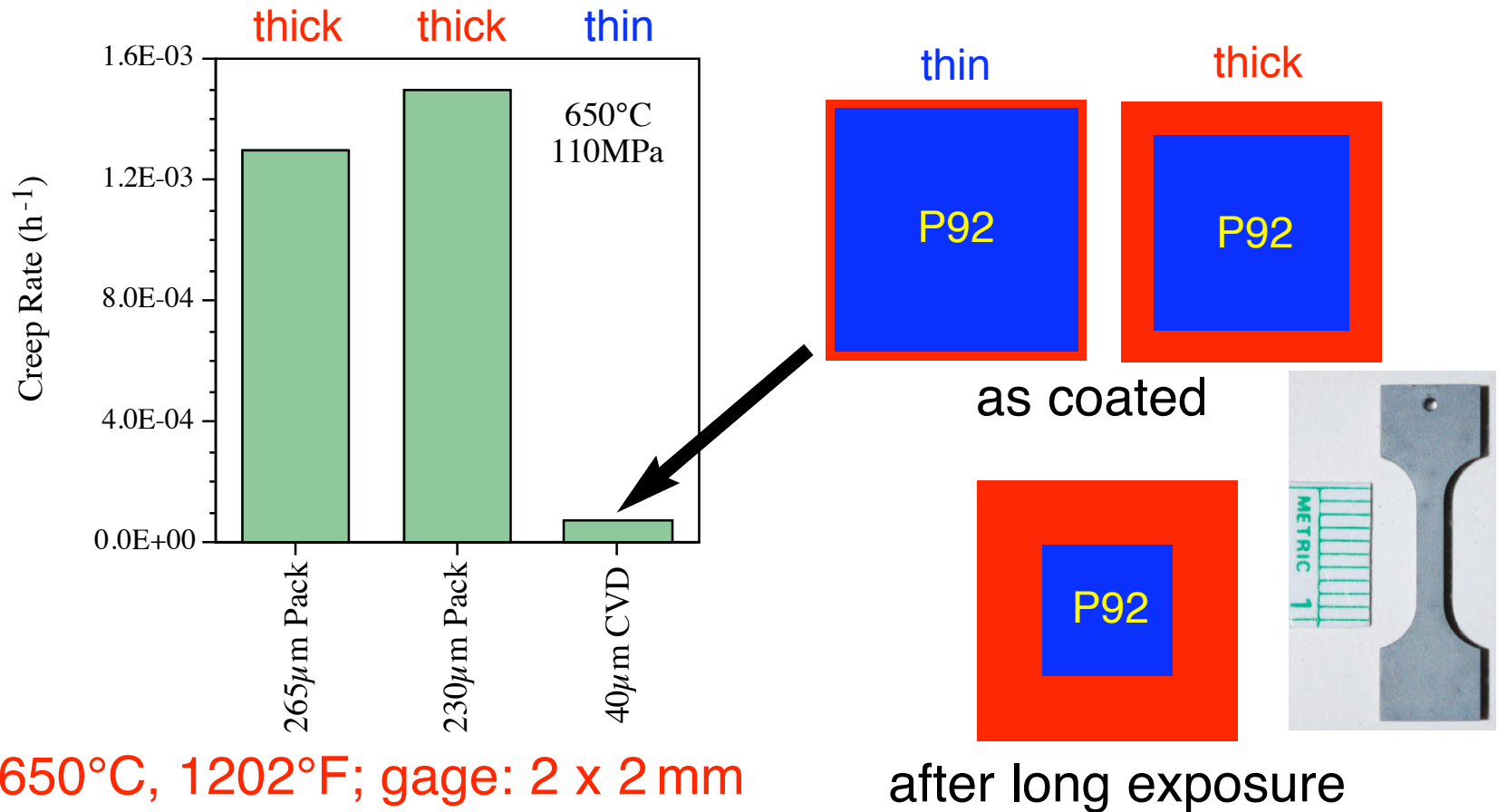
thermal expansion mismatch problem

Al loss: more by interdiffusion than oxidation

Al interdiffusion destroys Fe-Cr strength

Creep Testing of P92 (Fe-9Cr-2W)

Effect of as-deposited coating thickness



Specimen with thin coating has better creep resistance
Effect of coating can be modeled as if coated layer absent

Suggests that thin coatings are preferable

Dryepondt et al., Surf. Coat. Tech. (2006)

Key Points

Normally think that more Al is better

thermal expansion mismatch problem

Al loss: more by interdiffusion than oxidation

Al interdiffusion destroys Fe-Cr strength

less is more: “thin” $\sim 50\mu\text{m}$ coatings better

Key Points

Normally think that more Al is better
thermal expansion mismatch problem

Al loss: more by interdiffusion than oxidation

Al interdiffusion destroys Fe-Cr strength

less is more: “thin” $\sim 50\mu\text{m}$ coatings better

T91 lifetime model developed

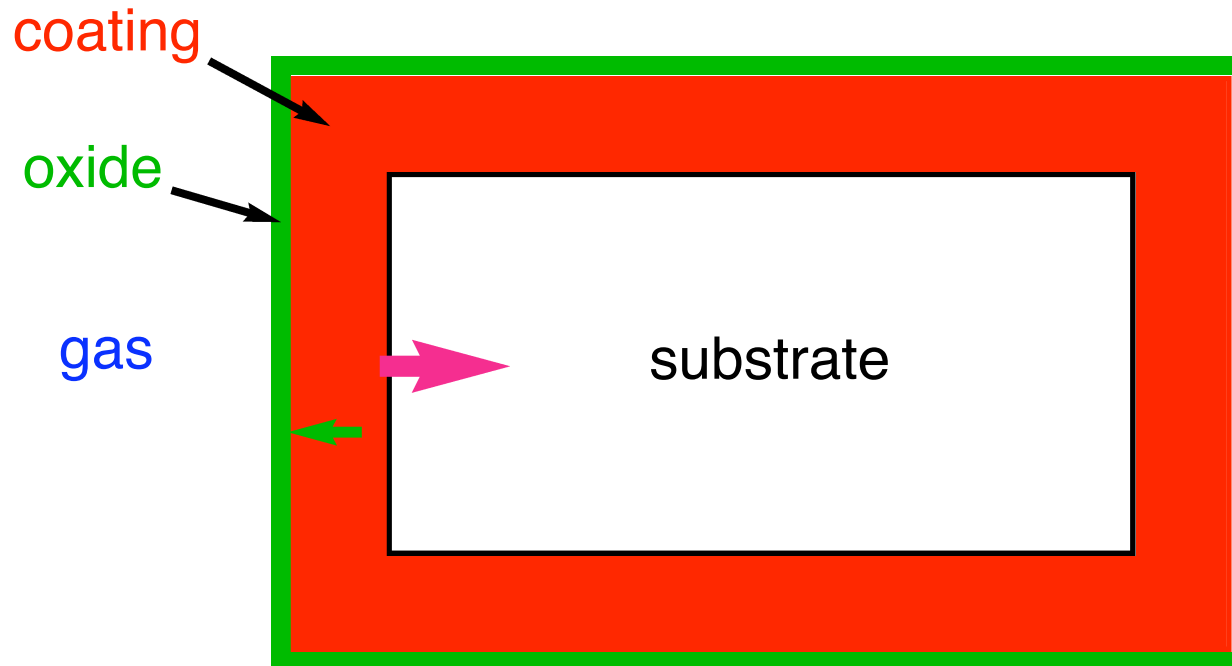
Based on NASA COSIM model (Nesbitt)

Isothermal diffusion experiment data used

Need failure criteria to predict life

Defining a coating failure criteria

need to determine C_b for coating in steam



Al supply: coating thickness and starting Al concentration

Coating thickness loss or Al content drop due to:

- (1) oxidation/sulfidation: selective formation of reaction product
- (2) diffusion into substrate

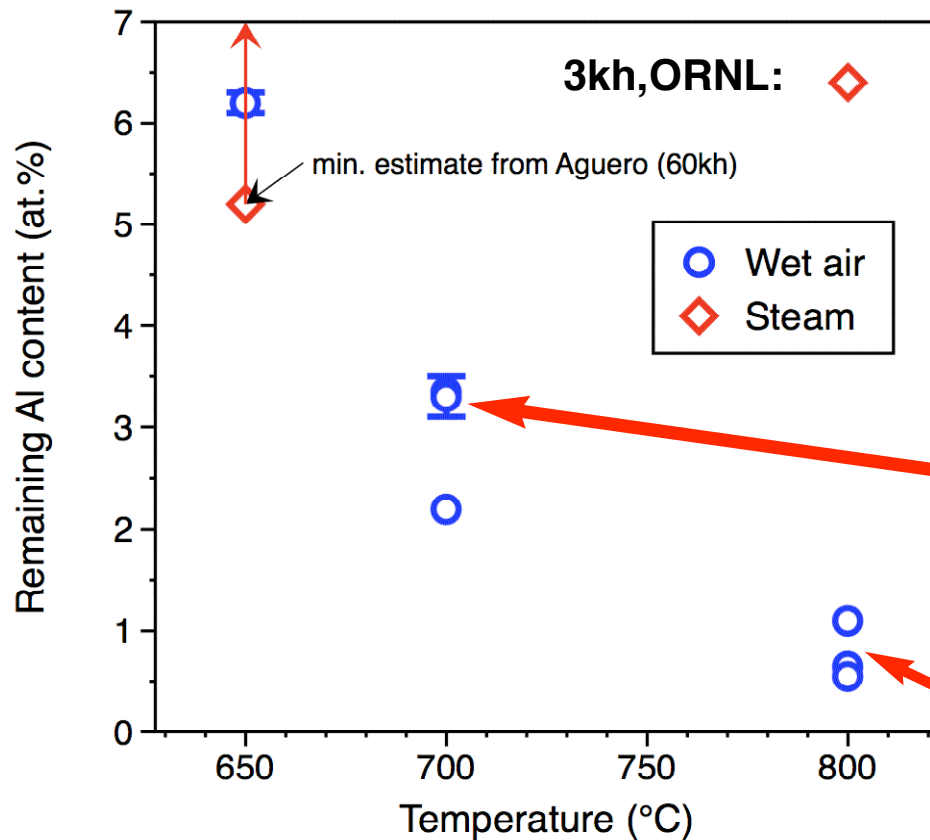
At low temperatures 650-700°C expect loss by (1) \ll loss by (2)

$C_b \sim 20\text{at\%Al}$ for sulfidation

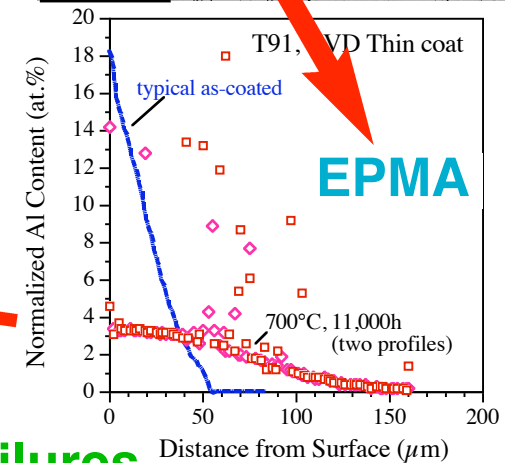
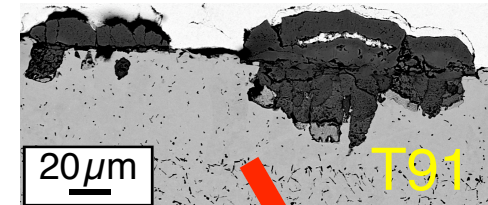
How low can Al content drop in steam environment?

Effect of temperature on C_b

~40 μm coatings on Fe-Cr at 650°-800°C in H₂O



coating failure



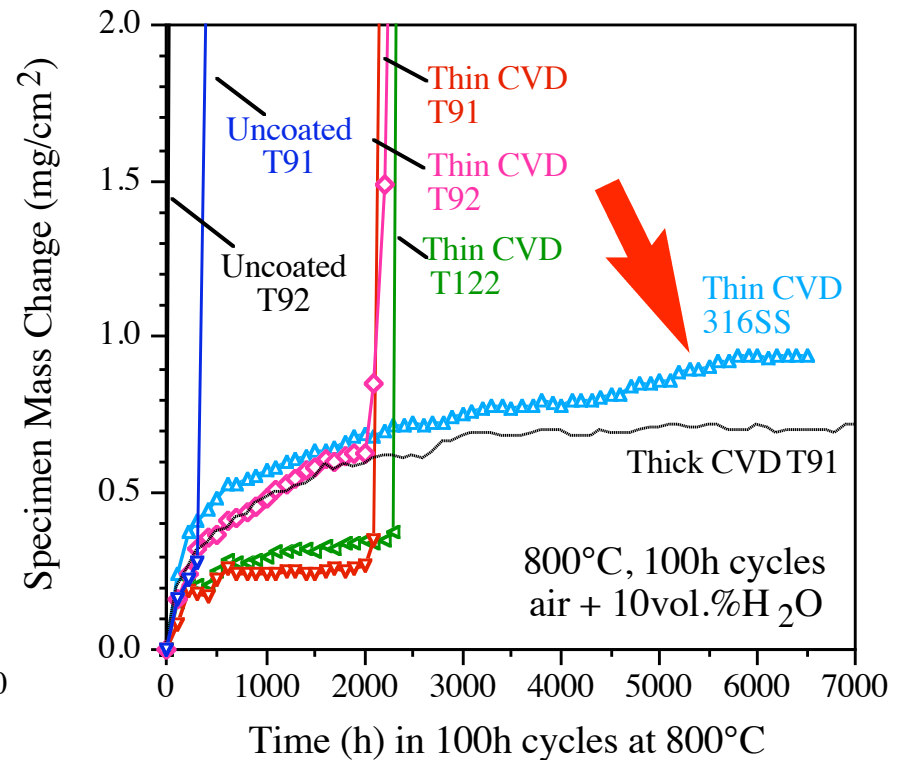
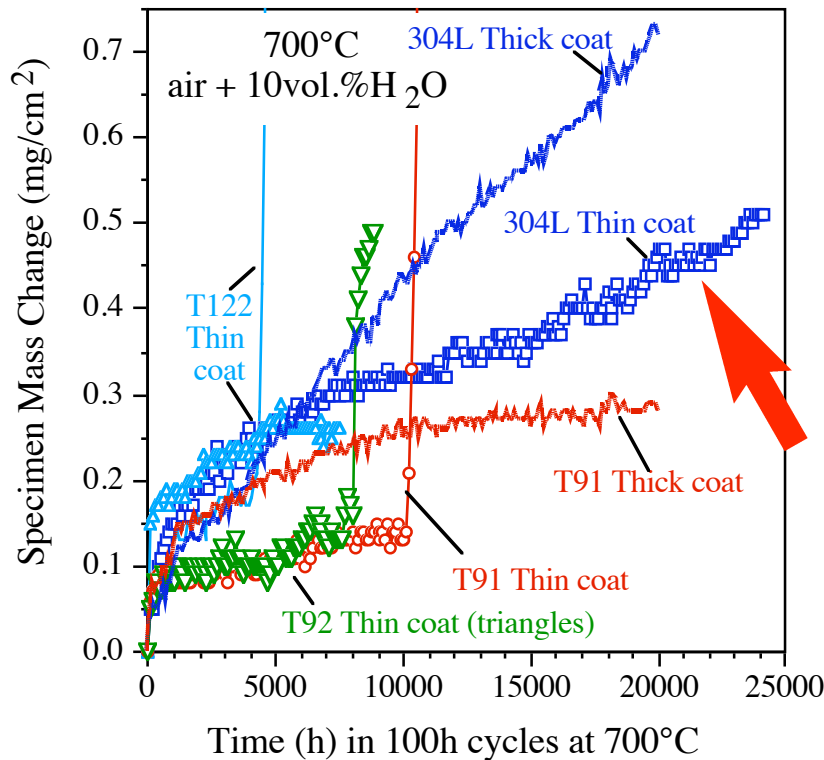
3 800°C failures
T91, T92, T122
(Fe-8-11%Cr)

Six failures of thin coatings, one higher Al activity coating
Agüero: 650°C slurry coating failed at ~60kh in steam

If temperature relationship is understood, this data set
forms the basis for a comprehensive lifetime model

No austenitic failures!

Thin coatings at 700° and 800°C in wet air



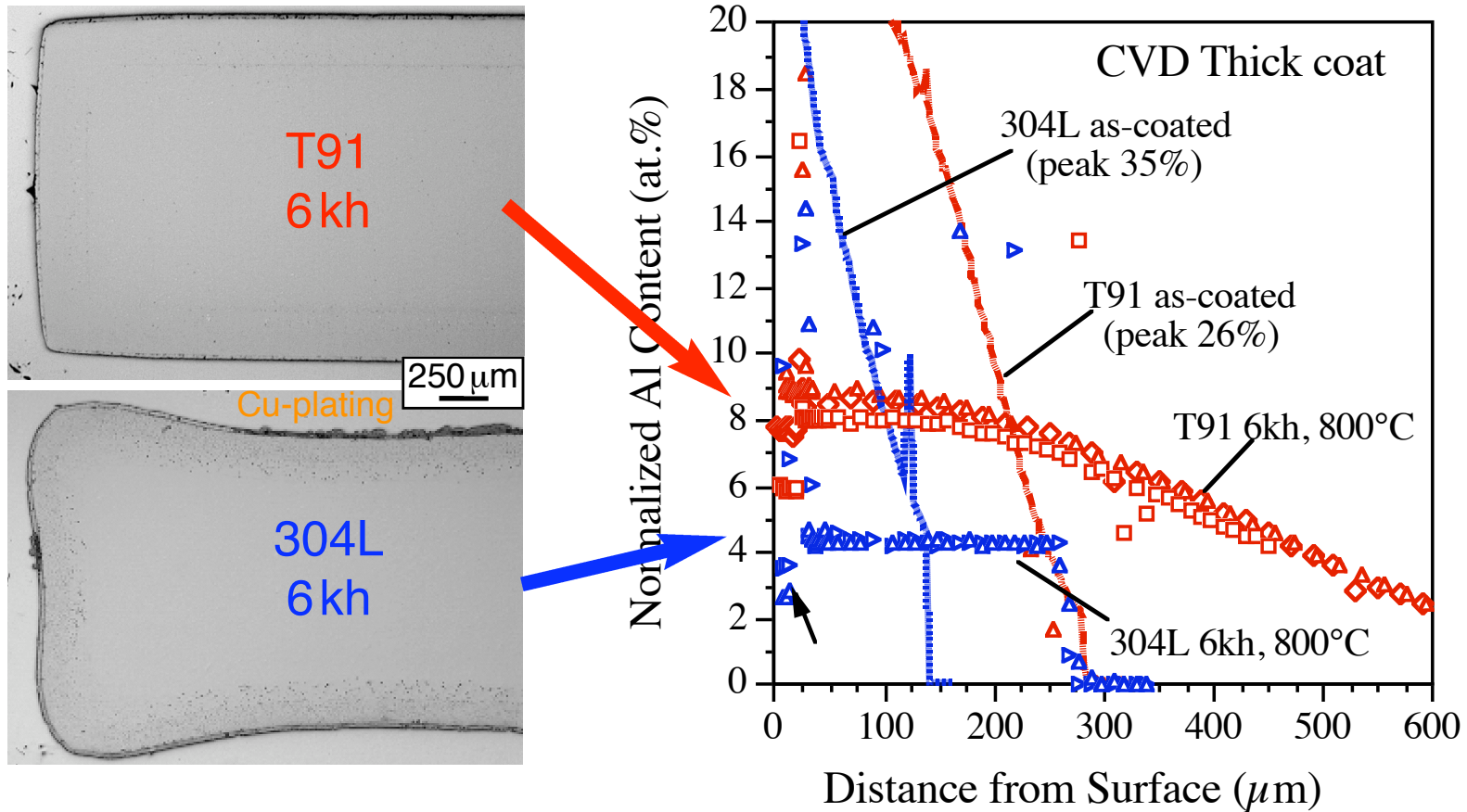
700°C: thin coating on 304L past 24 kh
>2X coating life on T91

800°C: thin coating on 316SS past 6 kh
almost 3X coating life on T91, T92, T122

Original hypothesis: Higher Cr content in coating (~18%Cr)?

800°C interdiffusion difference

250 μ m coatings stopped 6kh without failure



Austenite-ferrite phase boundary
inhibits Al interdiffusion

Key Points

Normally think that more Al is better
thermal expansion mismatch problem
Al loss: interdiffusion more than oxidation
less is more: “thin” $\sim 50\mu\text{m}$ coatings better

T91 Lifetime model developed

Based on NASA COSIM model (Nesbitt)
Isothermal diffusion experiment data used
Need failure criteria to predict life
8-10%Cr doesn't affect lifetime at 800°C
 C_b decreases with increasing temperature
No failures on thin coatings on 304L/316
Model still being developed for 304L
how account for phase boundary effect?

Coatings: A path forward

- Coupon tests only of interest if taken to failure
 - Reality for T91 (all F/M):
 - boiler application $\leq 600^{\circ}\text{C}$ - no Al interdiffusion
 - Thin coating will minimize mech. problems
 - Reality for 304L (347HFG, Super304H, etc.)
 - no boiler application above 650°C
 - phase boundary will limit interdiffusion
 - Will CTE mismatch cause fatigue cracking?**
 - Fabrication
 - How to coat 10m tubes? **Thin slurry coating?**
 - How to coat welds? (Coat before PWHT!)
 - Demonstration
 - 304/347 tube explosions creates need
- Now: 2010 EPRI program with ORNL & Praxair**

Milestones

FY09: Coatings

Done - Complete 800°C testing (9-12Cr failures)

Done - Final report on coatings: presentation at Oct. 09 EFC workshop - 2010 proceedings

Future: EPRI demonstration coating 347H tubes
Tritium permeation barriers

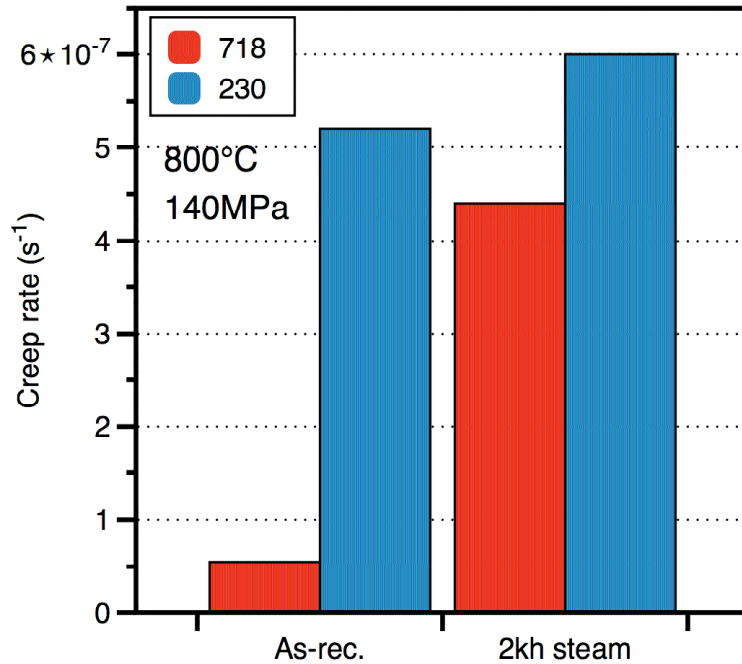
FY10: Oxy-firing corrosion

- Literature review
- Begin testing in CO₂, H₂O (i.e. no ash)
(piggy-backing on USC: 800°C, 17bar steam)
- Begin testing in fireside corrosion (**done**)
- Build in-situ rig for creep testing in steam

What is effect of steam on creep?

Little experience with new alloys in steam

ex-situ (in air)



in-situ



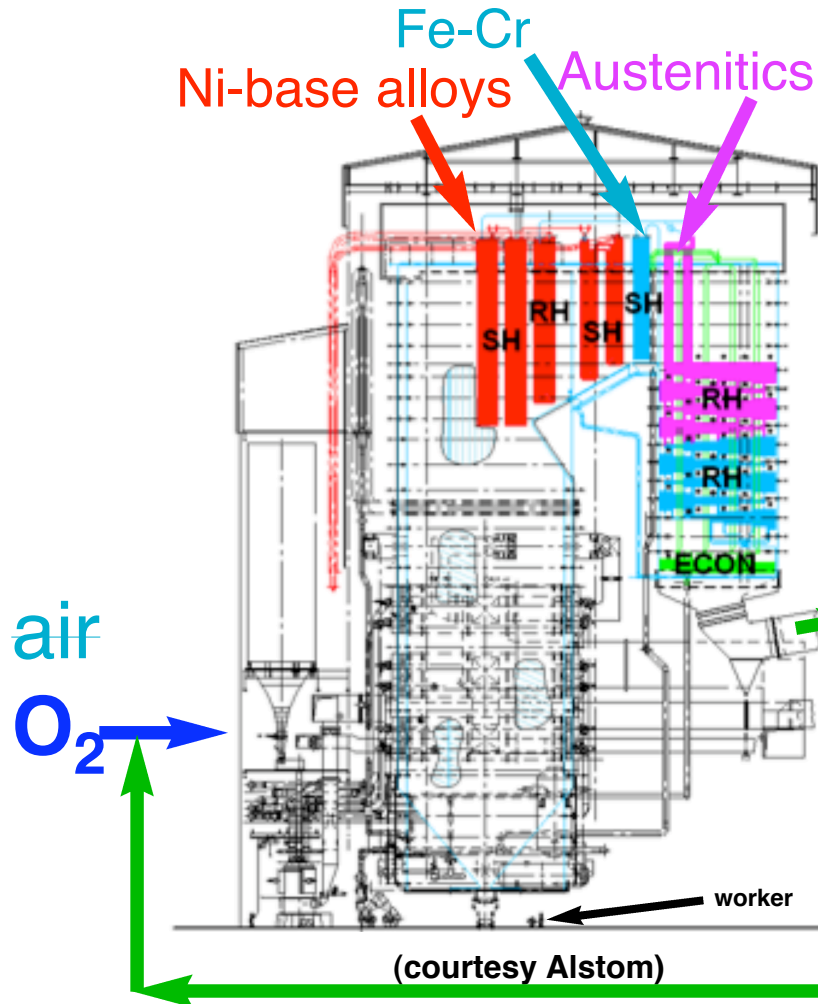
Untested: C617, 740
also 4 kh steam, 2 kh anneal

under construction

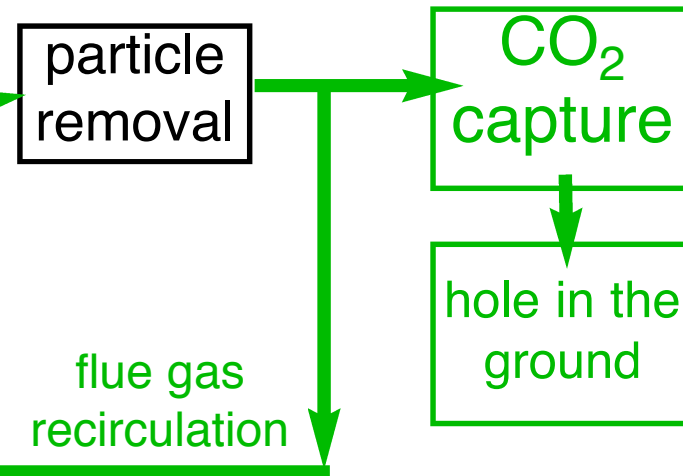
In H₂O, well-known that H injected into metal: effect?
Better comparison: coating debit vs. corrosion debit

Introduction - Oxy-fired Coal Boilers

Several studies published by Alstom (Bordenet)



	air	O ₂
CO ₂	15	59%
H ₂ O	10	32%
O ₂	2.5	1.9%
SO ₂	0.13	0.46%

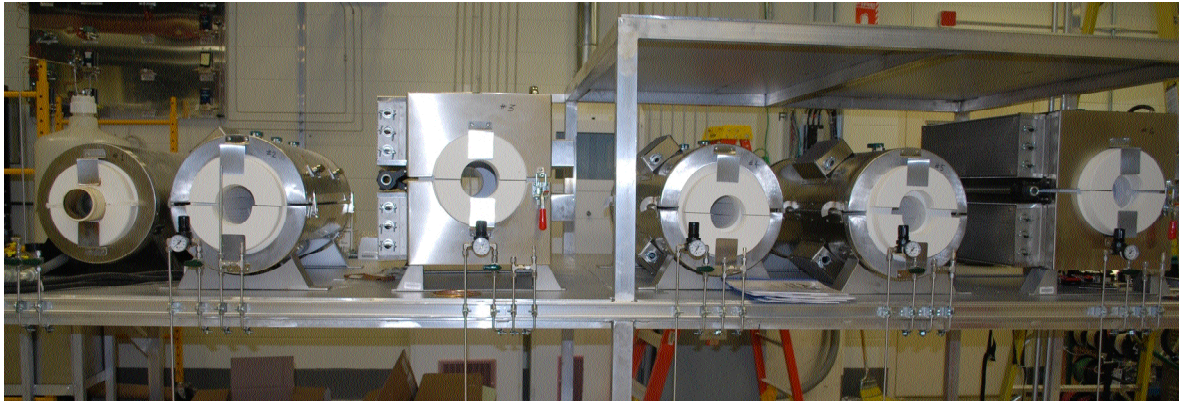


Germany: 30MW oxy-fired pilot plant (Alstom)

U.S. utilities: no oxy-firing without CO₂ legislation

Corrosion testing

Determine effect of higher CO₂, H₂O, SO₂...



gas only, no ash
- H₂O only
- CO₂ only
- H₂O-CO₂
(under construction)



Synthetic ash: 30%Fe₂O₃-30%Al₂O₃-
30%SiO₂-5%Na₂SO₄-5%K₂SO₄
Gas: N₂-15%CO₂-3.5%O₂-0.25SO₂
Temperature: 600°C
Time: 500h (1 cycle)

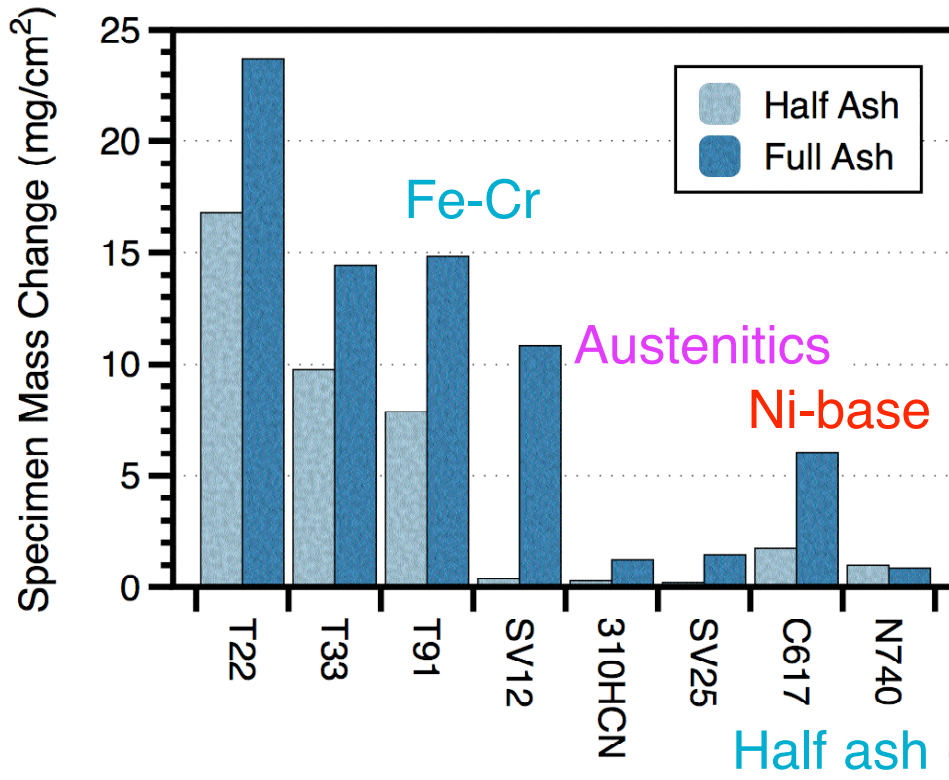


Porous alumina

Get started: establish methodology + safety/health
make specimens, work out experimental issues

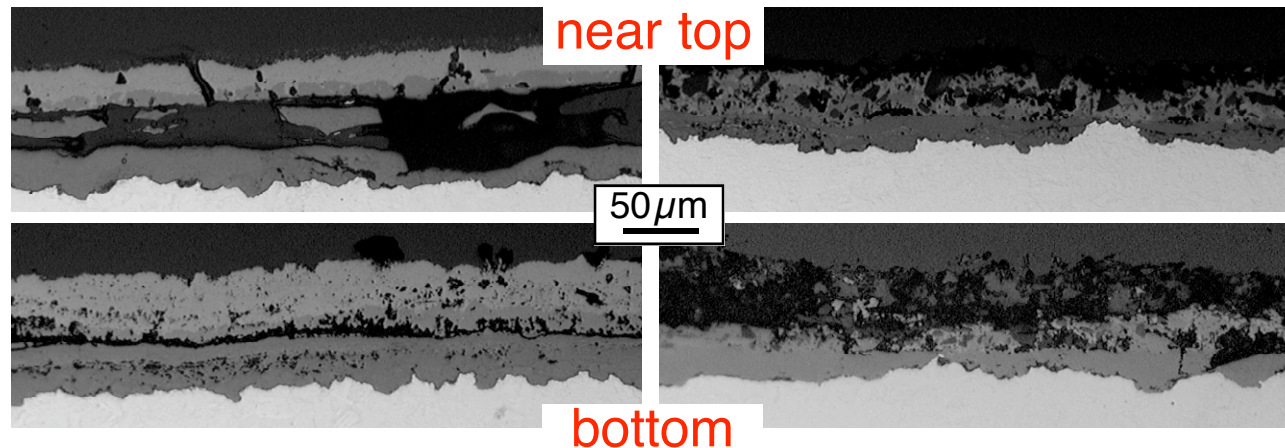
Half ash vs. full coverage

Different procedures reported in literature



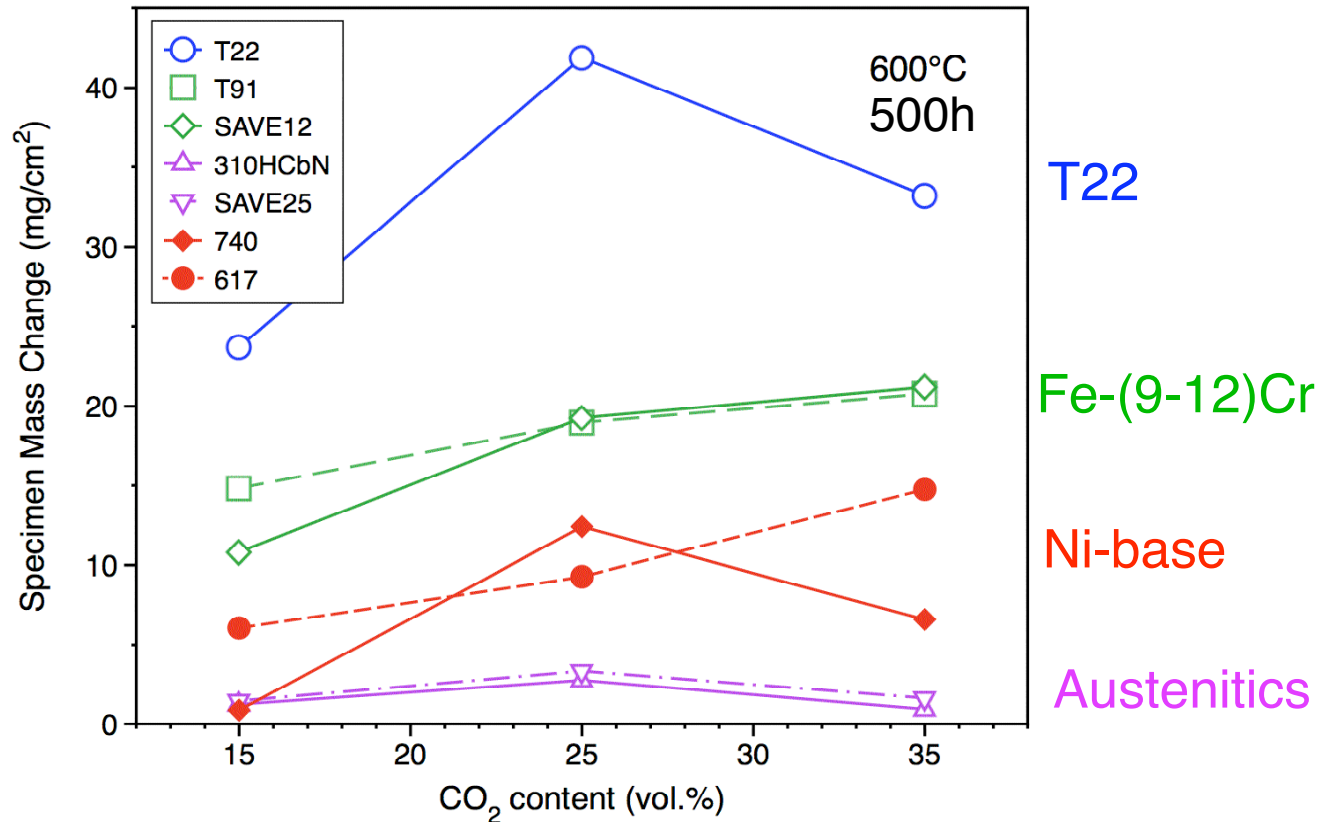
Half ash (3.8g)

Full ash (9g)



Preliminary results: ΔCO_2

Increased CO_2 concentration in base gas



Mass change data only - need metal loss
Qualitatively similar to 700°C Bordenet data
(austenitics better than Ni-base)

Not just commercial alloys

Model alloys: better composition understanding

		Ni Content (wt.%)			
		0	25	35	45
Cr Content (wt.%)	15	X			
	20	X			
	25		X	X	X
	35		X	X	X

Alloy 33

chromizing

622,625

Cast, hot-rolled to 8mm: cut coupons and rods

Potential coating compositions

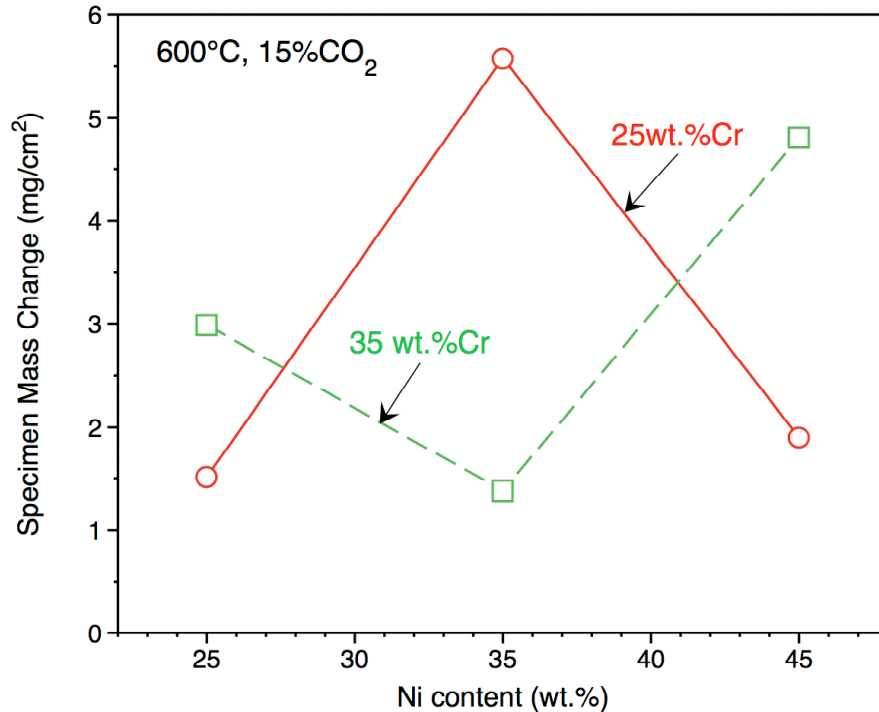
Information for alloy development

Future: quaternary additions, etc.

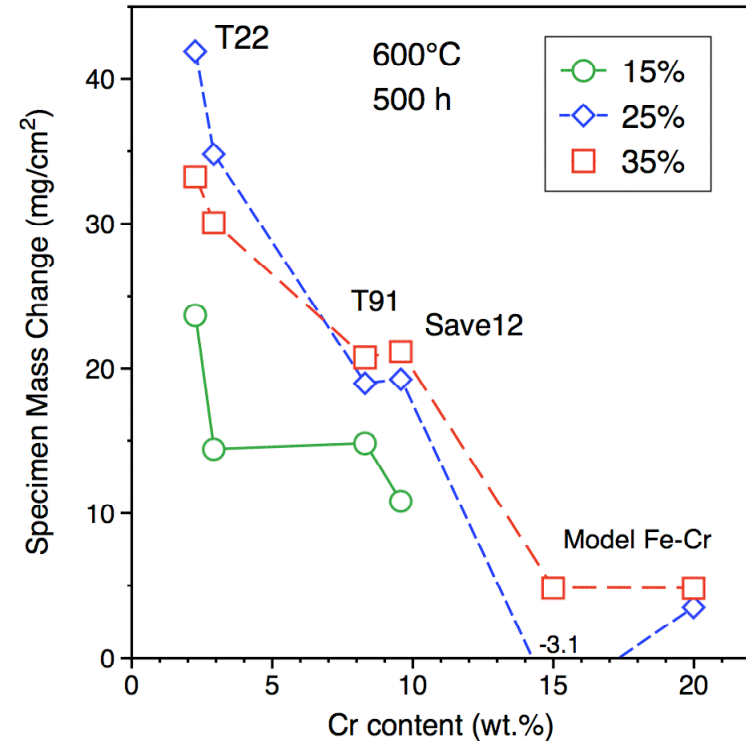
Initial model alloy results

Ash tests, 500h 600°C

FeCrNi alloys



Fe-Cr alloys



Mass change data only - need metal loss
35%Cr alloys significantly less pitting than 25%Cr

Long-term plan

FY10: Setup

- Literature review
- Begin testing in CO₂, H₂O (i.e. no ash)
- Begin testing in fireside corrosion (**done**)
- Build in-situ rig for creep testing in steam

FY11: Complete work on Fe-Cr

FY12: Austenitics

FY13: Ni-base alloys

backups

High-purity well-controlled process

Chemical vapor deposition (CVD) coatings

similar to a well-controlled above-pack coating process

ORNL laboratory scale reactor with 2-4, 2x1 cm specimens

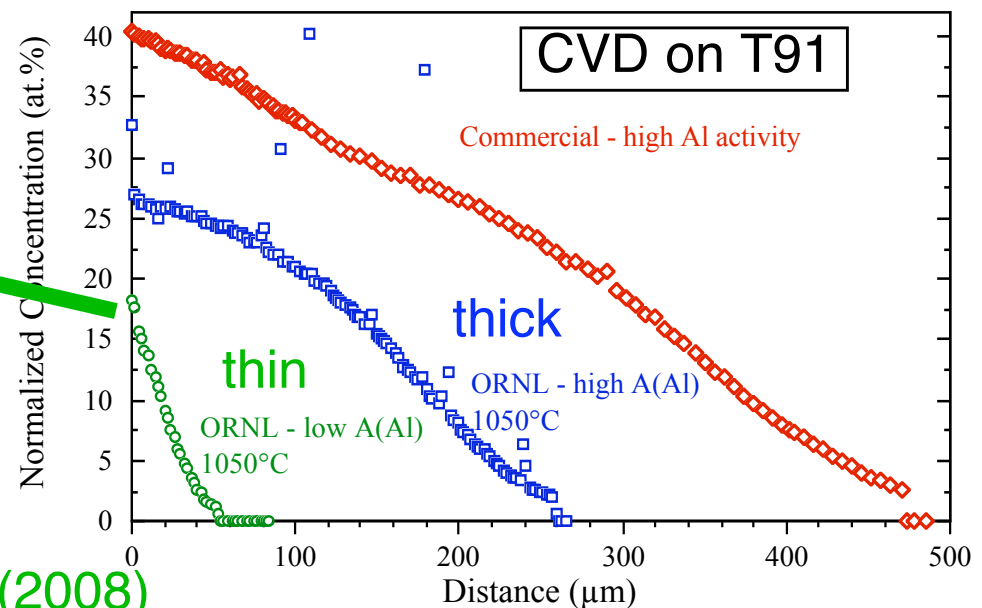
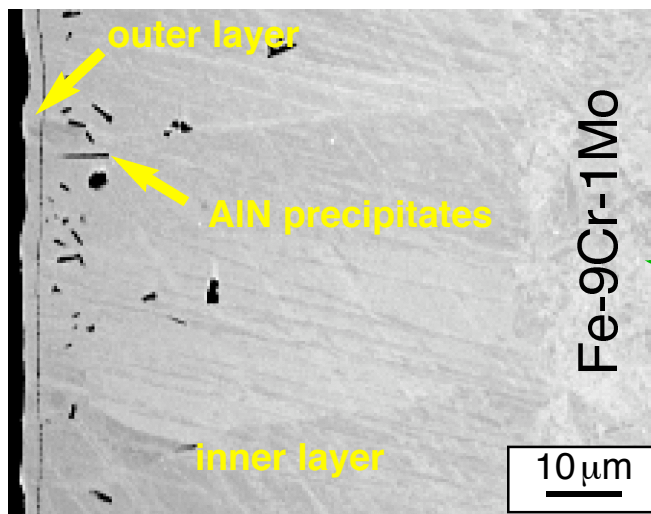
austenitic 304L (Fe-18Cr-8Ni)

ferritic-martensitic T91 (Fe-9Cr-1Mo)

flowing H_2-AlCl_x , 100 Torr, 6h, 900°C or 4-6h at 1050°C

Two types: “Thick” coatings $\approx 40\mu m$ Al-rich outer layer, 150 μm total

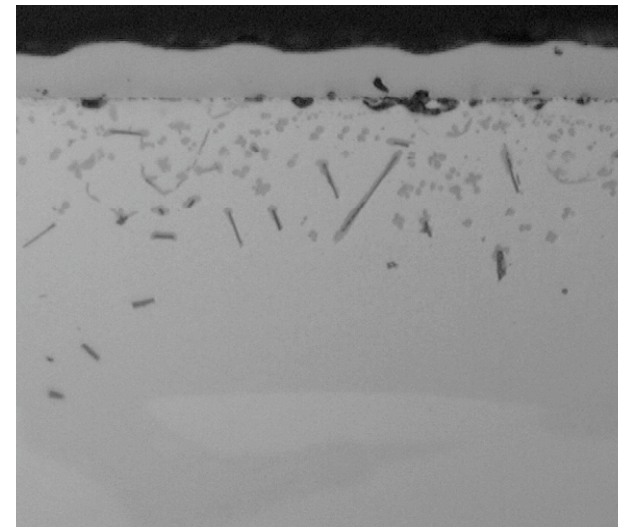
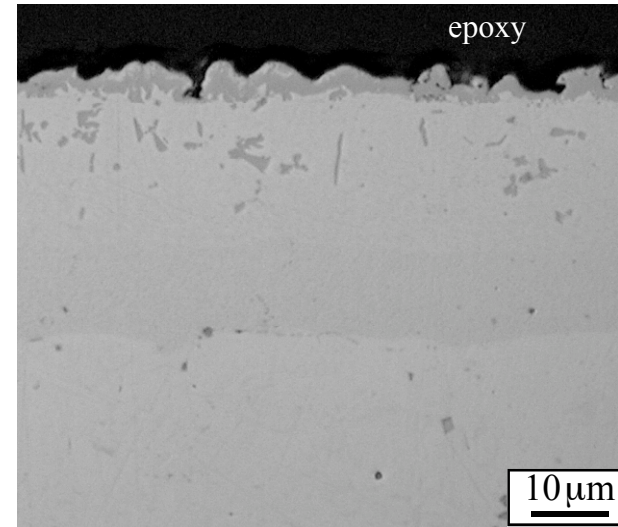
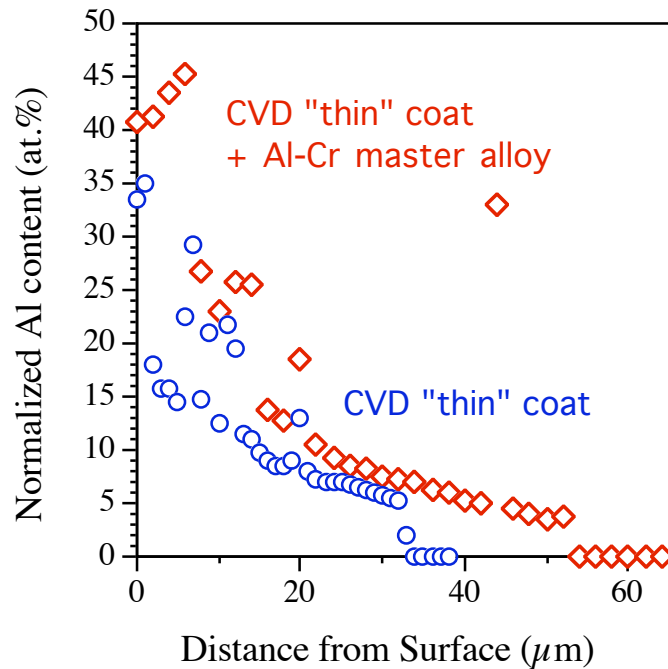
“Thin” $\approx 5\mu m$ Al-rich outer layer, 35 μm total thickness



Zhang et al., Surf. Coat. Tech. (2008)

Less thin coatings

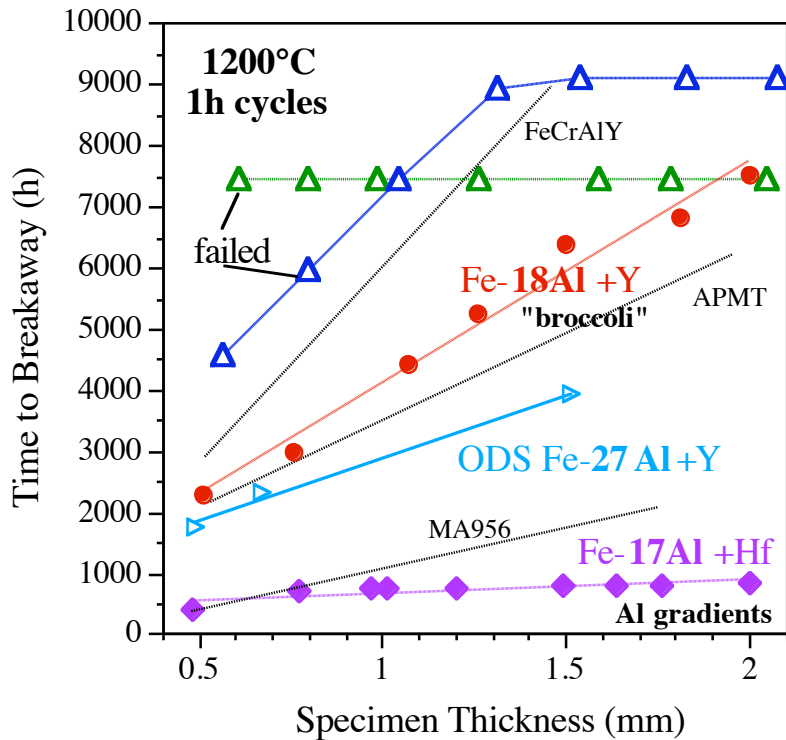
Modify CVD process by adding Al-Cr powder



Improved processing for H permeation barriers

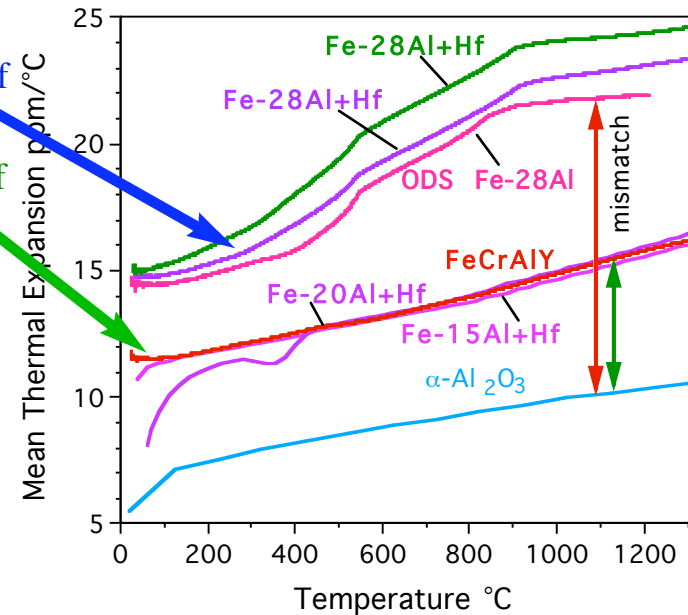
Final work: Fe-Al alloys + Y, Hf

Academic exercise to see comparison



Fe-28Al+Y/Hf
3 in progress

Fe-19Al+Y/Hf
6 in progress



DO₃-> <- B2 -> <- α

Phase change with T

Fe-17Al+Hf: Al gradients reduce life

Fe-18Al+Y: broccoli failure (shorter life than FeCrAlY)

Fe-19Al+Y/Hf: Optimized dopant, highest Fe(Al)

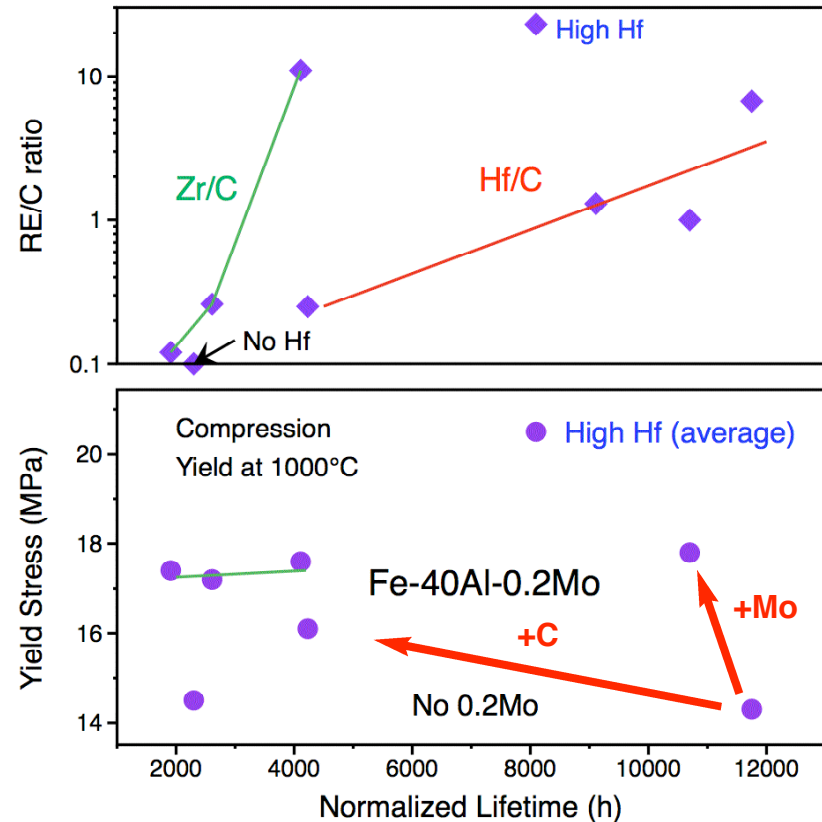
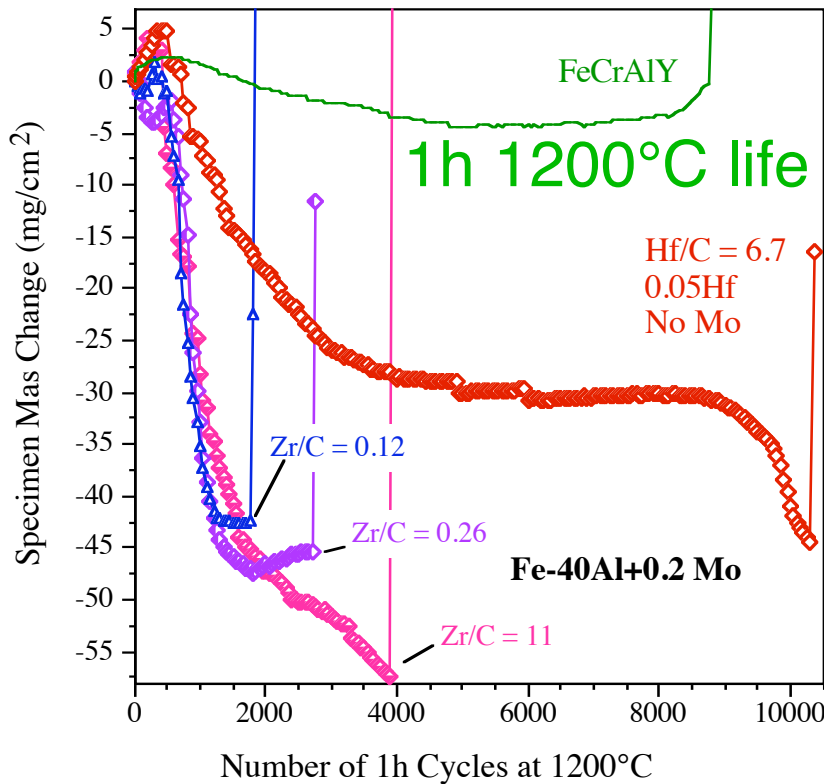
Fe-28Al-2Cr+Y/Hf: Fe₃Al (high CTE)

5kh/yr: taking a long time to complete!

Case study: Fe-40Al doping

10+ year argument on RE strengthening

ORNL base alloy: Fe-40Al-0.2Mo-0.05Zr-0.4C $Zr/C=0.12$



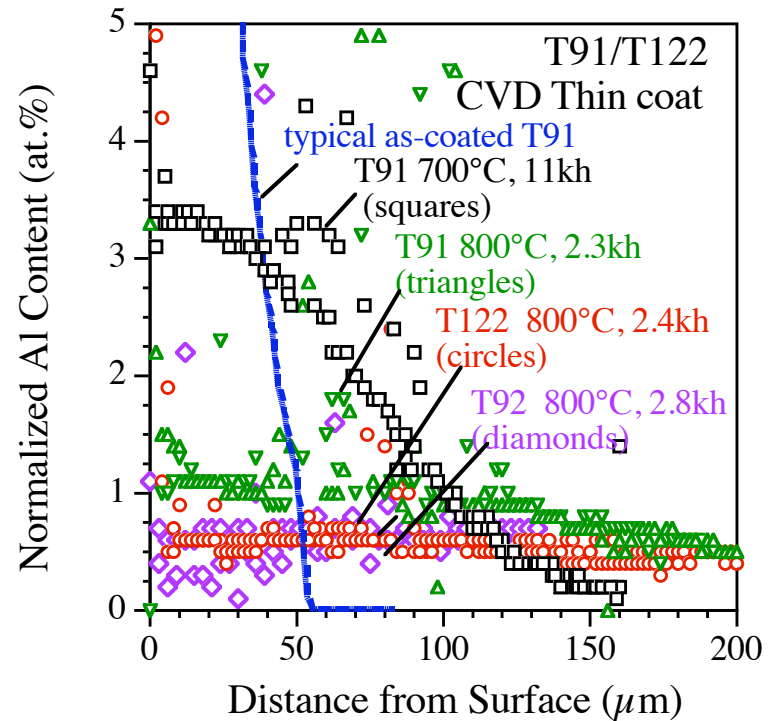
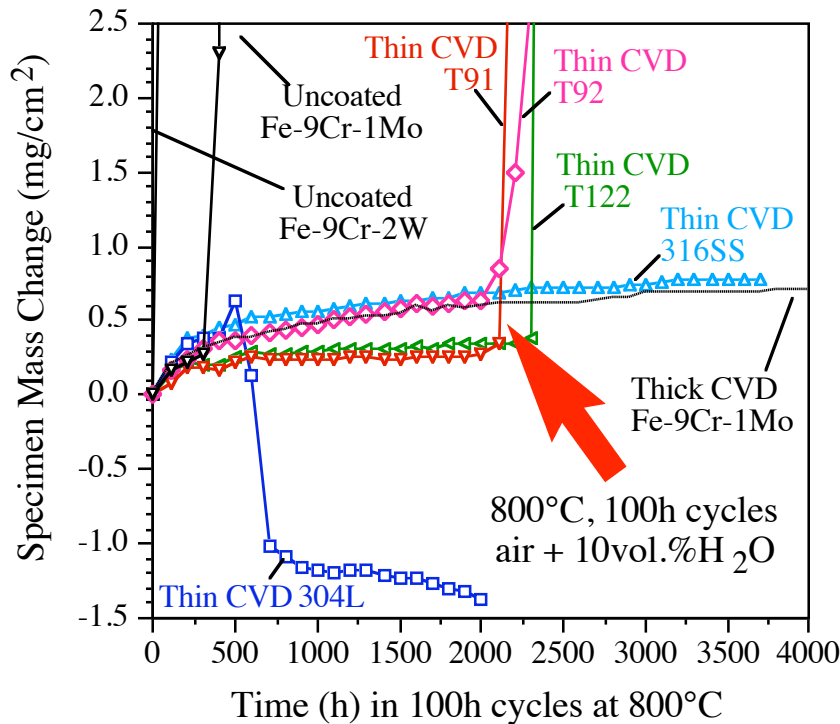
Whittenberger: in NiAl, Zr strengthening gone by 1100K

Only $1000^\circ C$ compression yield data (Schneibel, Dryepondt)

Fe-Al: RE/C predicts life better than yield strength

Effect of C_{Cr} on C_b

Thin coatings at 800°C in wet air



T91, T92, T122 (10.5%Cr) all failed within 10% of life

C_b at failure was <1.1 at% Al

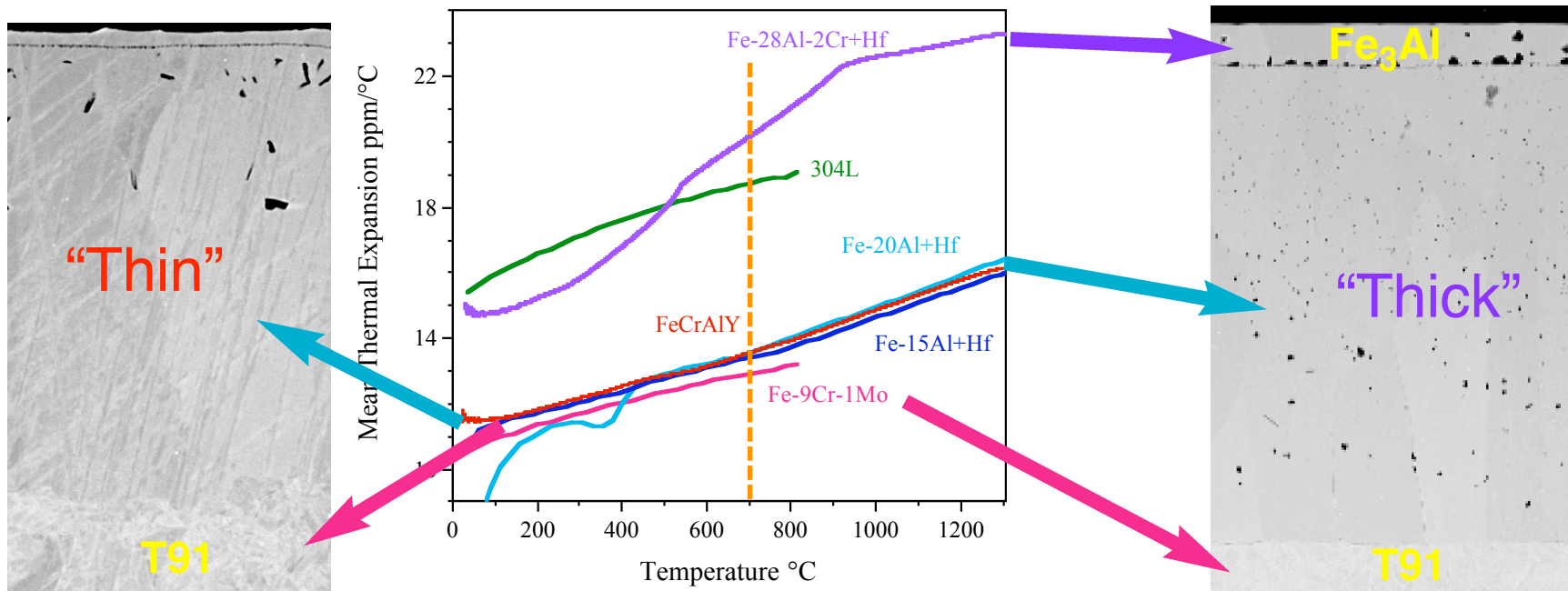
Kvernes (Fe-13Cr-xAl) said more Al required with T!

304L: bad coating

316SS: no failure at 4kh!

Fe-Cr Coating Solution

Eliminate CTE mismatch problem



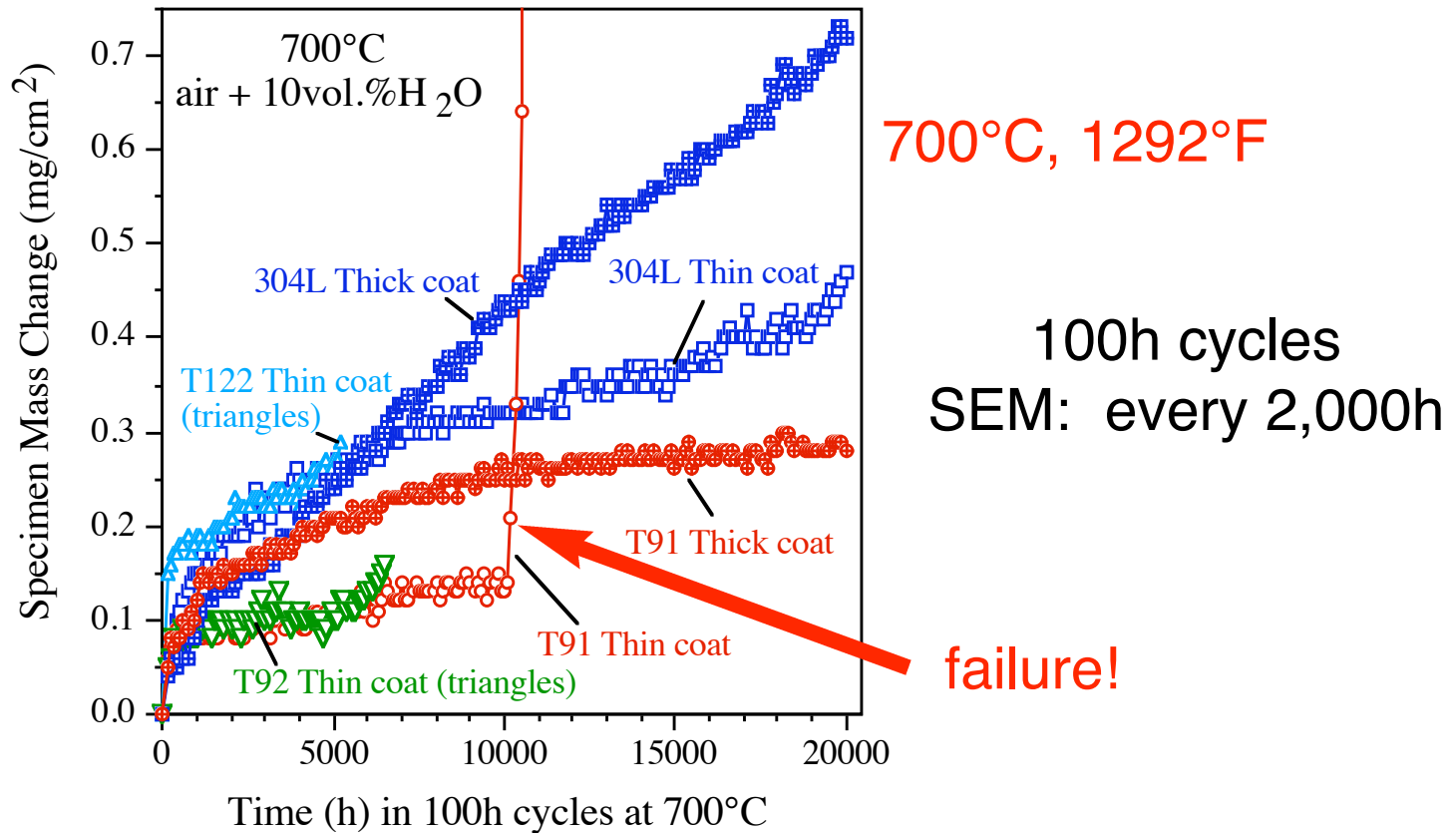
For coated T91:

Thick CVD: Outer Fe₃Al layer
inner coating & substrate are ferritic

Thin CVD: ~18at.%Al peak surface Al (no aluminide)
only α -Fe(Al) phase
NO Δ CTE

Coating Performance: 700°C

Accelerate failure by increased interdiffusion

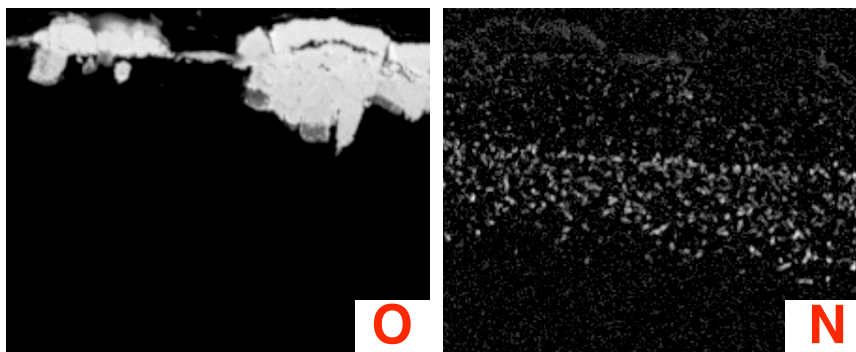
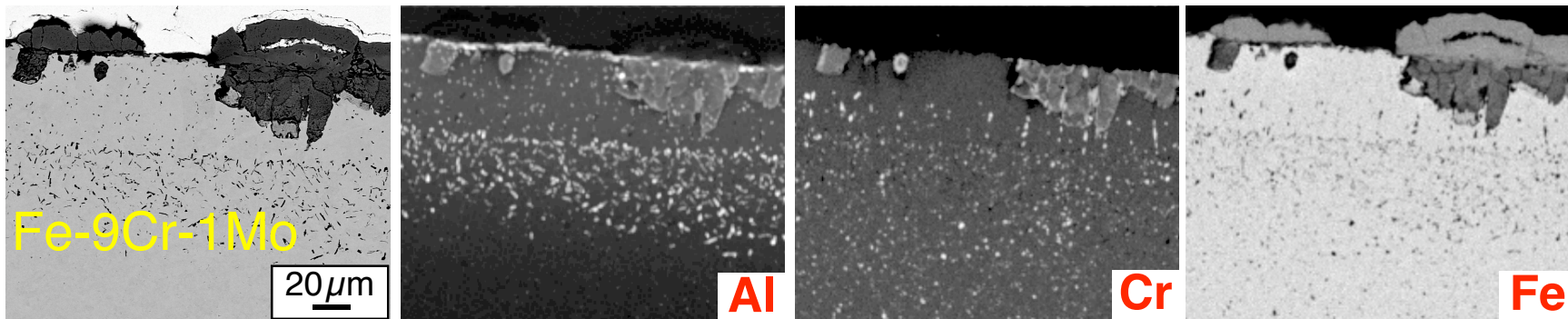


Thick coatings had a longer lifetime on T91 and 304L

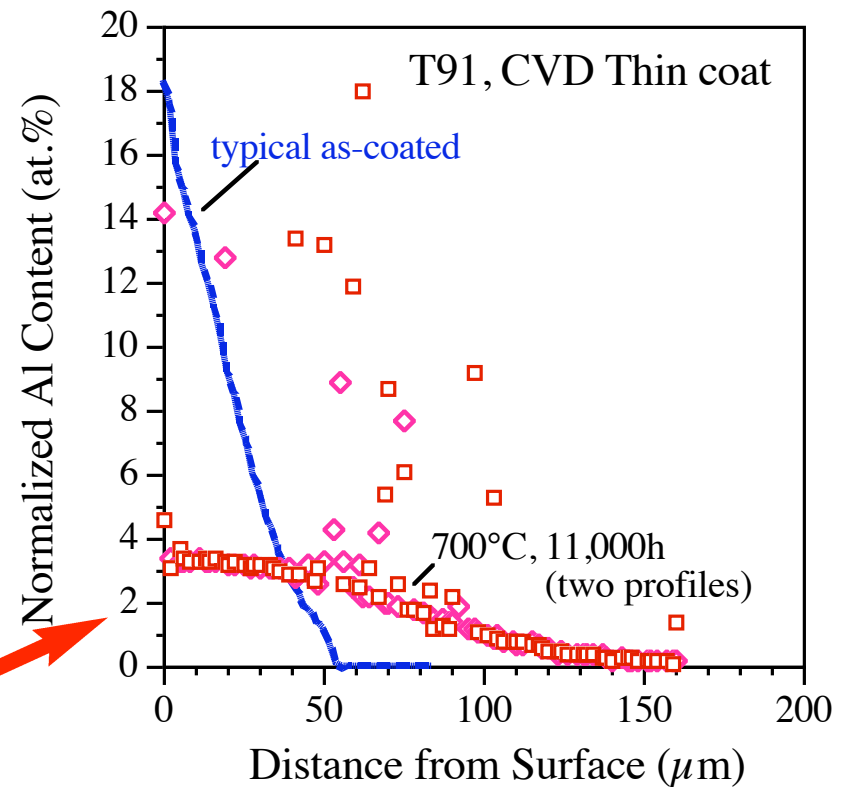
Breakaway oxidation for **thin coating on T91** at ~10,500h

Failed Coating Characterized

Thin coating after 11,000h in humid air at 700°C



Electron probe analysis:
Typical Fe-rich oxide nodules
AlN precipitates (0.2at.%N in alloy)
~3.4 at%Al at surface (C_b)



Diffusion predictions at 700°C

Using 10kh observations for $\sim 250\mu\text{m}$ coatings on T91

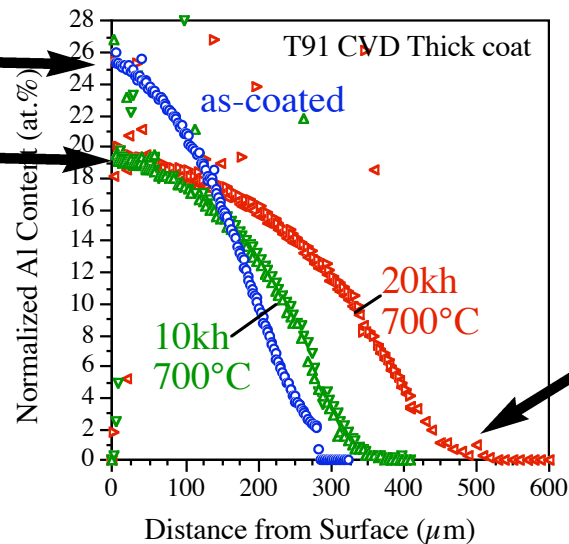
Prediction Method	COSIM dependent	COSIM indep. FeAl	COSIM indep. FeCrAl	Heckel	Actual (diffusion test)
Surface (at.%)	19	16	17	19	18%Al
Thickness (μm)	310	438	428	356	320 μm

COSIM - NASA (Nesbitt) developed model - missing key D_{xy} terms

Heckel - simple parabolic rate constant from 2,000h data

starting Al surface content

after 10kh diffusion at 700°C

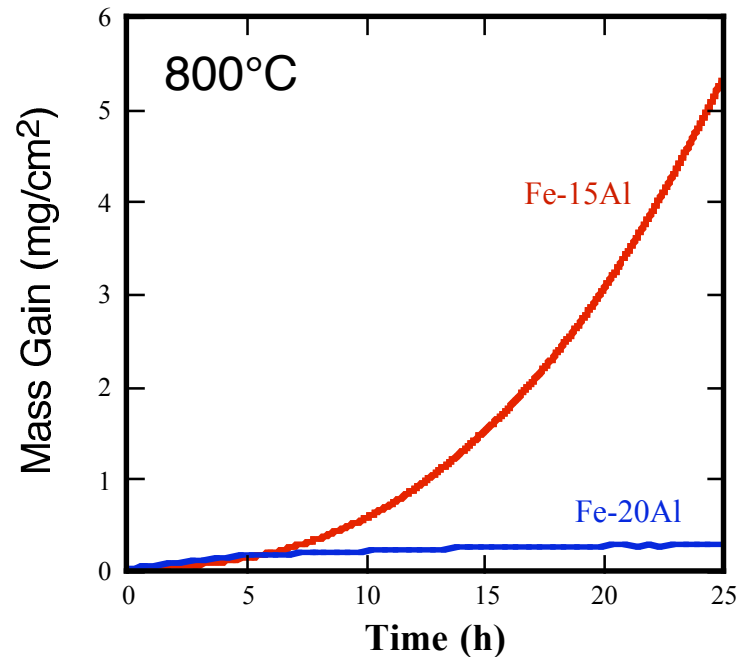


diffusion depth increases with t

Thin vs. Thick Coatings: Sulfidation

Model Fe-Al alloys at 800°C in H₂-H₂O-H₂S-Ar

800°C, 1472°F



For comparison to coatings, cast Fe-15at%(8wt%)Al & Fe-20%(11wt%)Al
Fe-15at%Al showed accelerated mass gain in test similar to thin coatings

Low mass gain for Fe-20%Al

Previous work by DeVan and Tortorelli found 18at%Al needed in Fe-Al

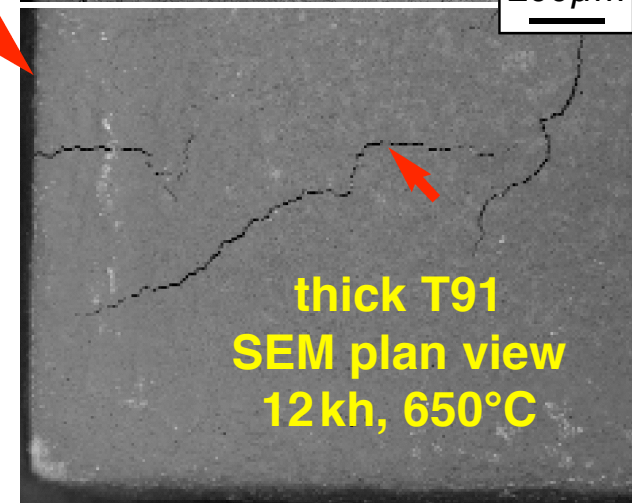
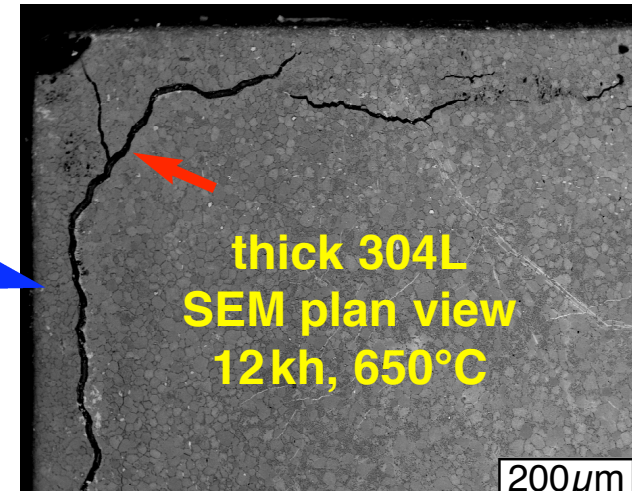
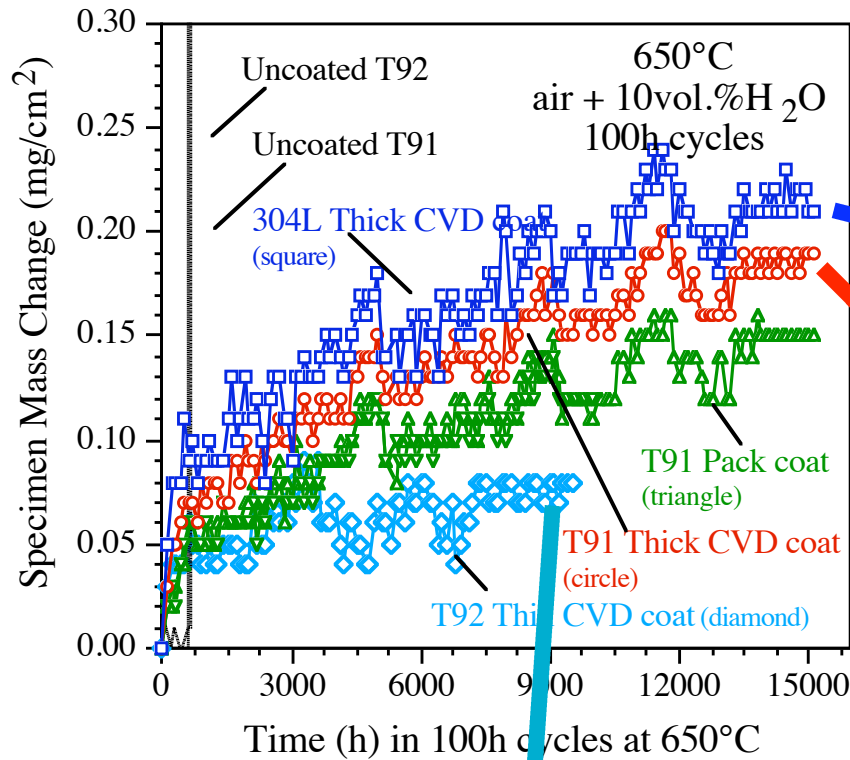
Higher Al content (“thick”) needed for sulfidation resistance

Summary

	Fe-9Cr	304L
CTE/cracking	↑	↓
Cr content	↓	↑
Phase boundary	↓	↑

Coating Performance: 650°C

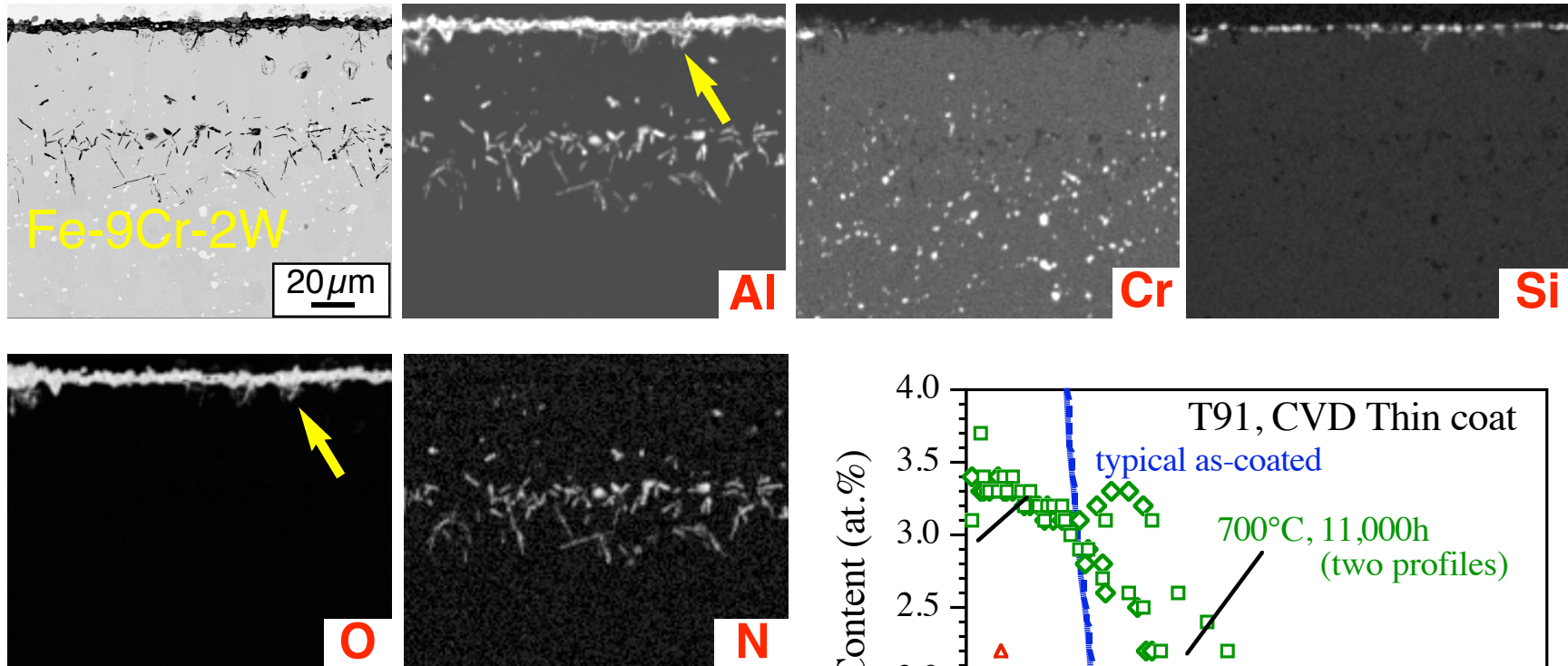
CVD or pack cementation coatings, 100h cycles



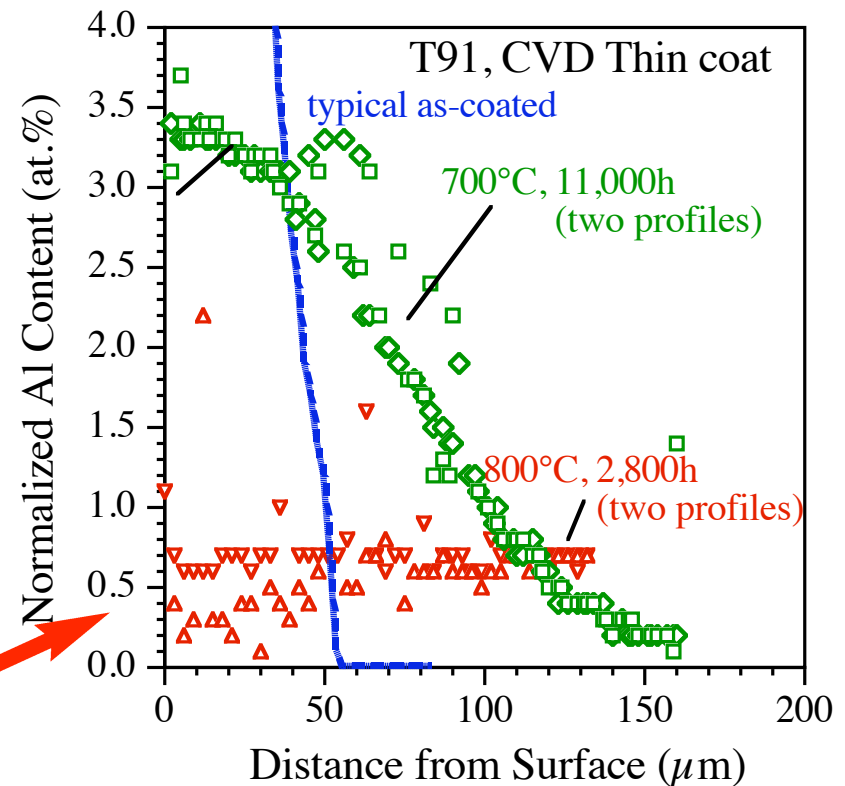
Not a fair comparison
Cracks appeared at ~10kh

Failed Coating at 800°C

Thin coating on P92 after 2,800h in humid air

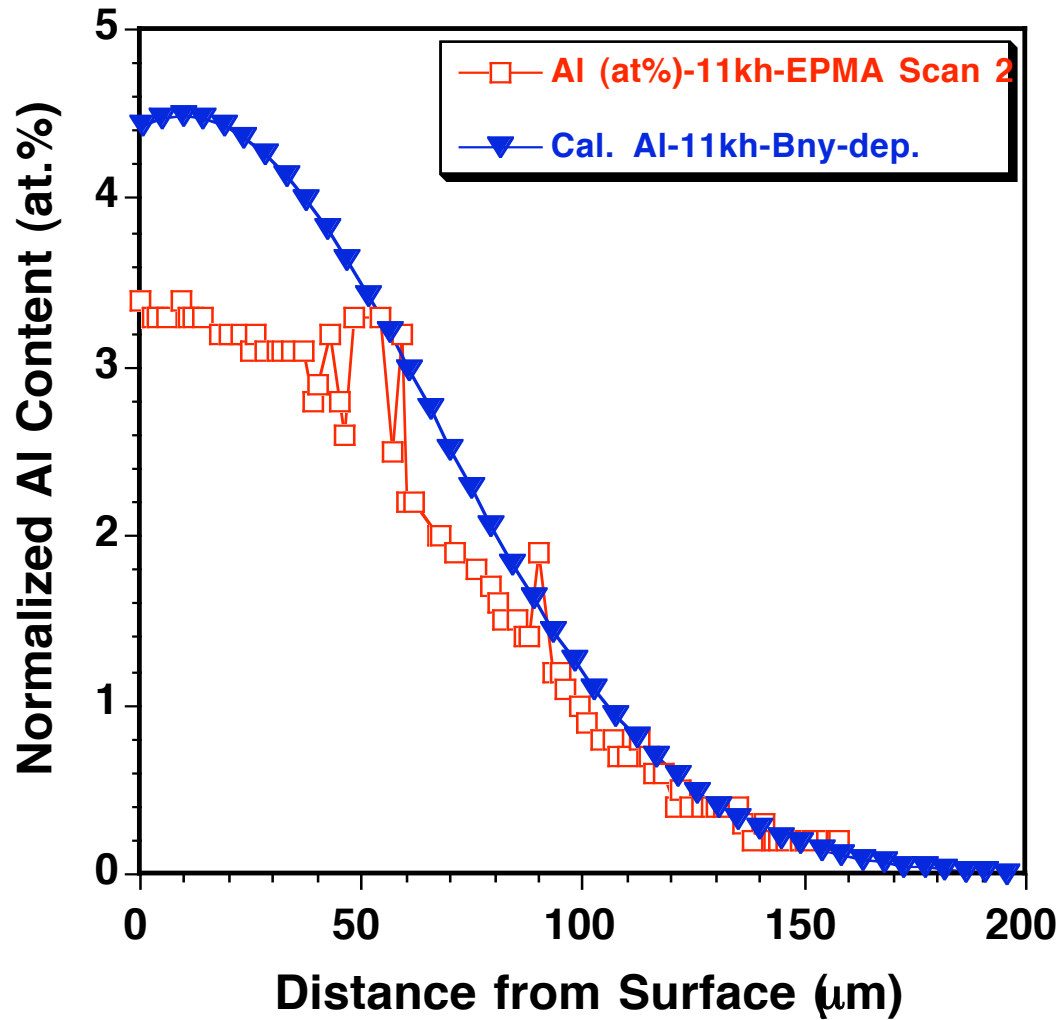


Electron probe analysis:
Area away from FeO nodules
Large AlN precipitates (N in alloy)
~0.7at%Al near surface (C_b)



T91 model for thin coatings

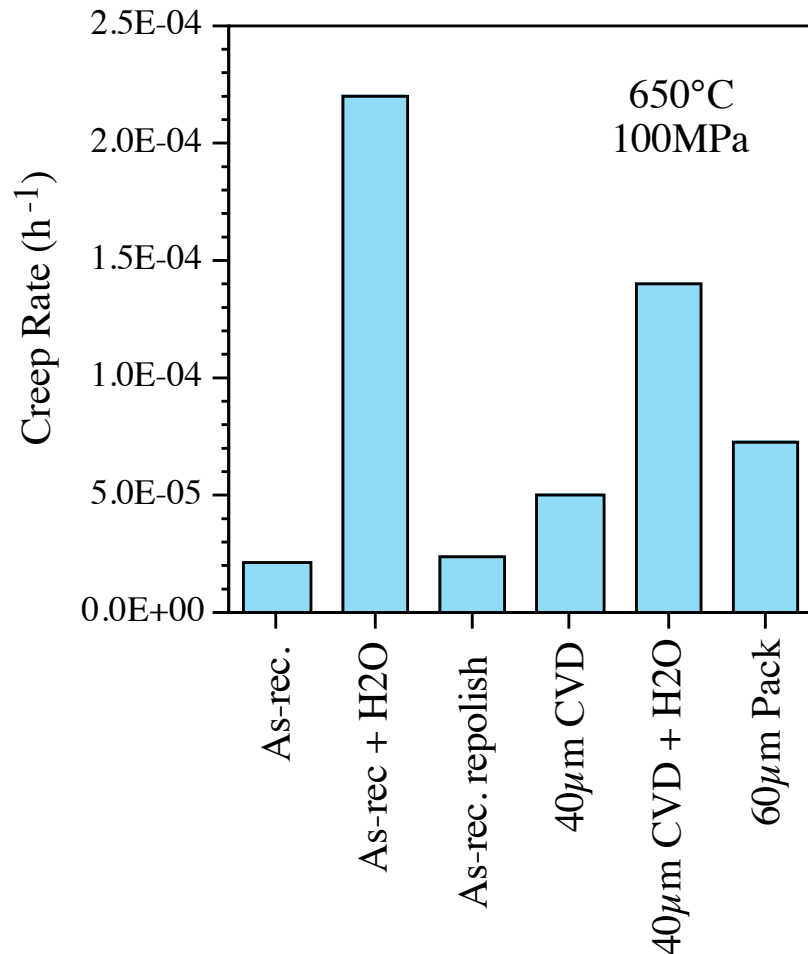
COSIM prediction based on 11kh failure



Need to refine Al consumption term to match profile

Combined Creep/Corrosion Testing

P92: **alternating** wet air & creep in dry air 650°C



650°C, 1202°F
100MPa, 14.5ksi

Ideally wanted to compare:

- loss of strength (coating)
40μm CVD, 60μm Pack
- uncoated loss (oxidation)

Order of testing effect:

creep test first - low rate
wet air test first -

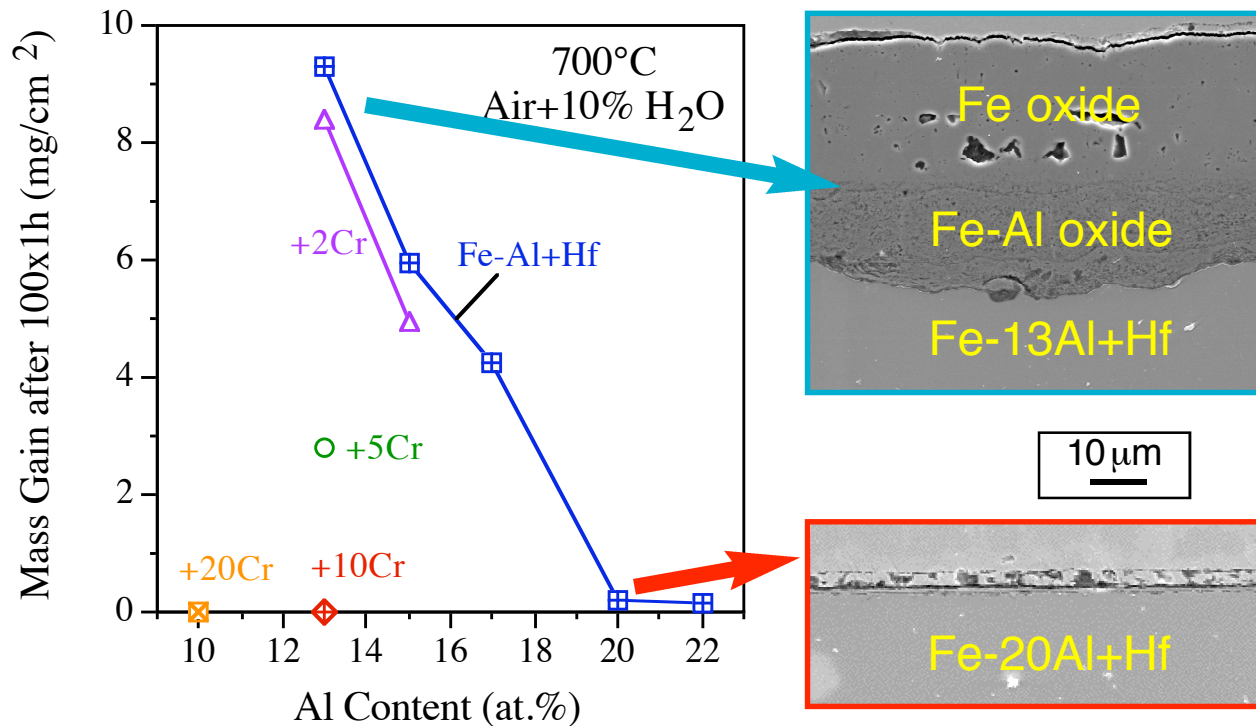
higher creep rate

Observations: uncoated specimen did not lose strength
>7% deformation did not breach coating (true for Fe₂Al₅?)

Creep testing in steam (or wet air) needed

Model Fe-Al-Cr alloy performance

after 100, 1h cycles at 700°C in humid air



Model alloys indicate with binary Fe-Al, need 20 at.%Al to avoid FeO_x formation

However, when Cr present, critical Al content drops with **10%Cr** (≈Fe-9Cr-1Mo), protective with 13%Al with **20%Cr** (≈304L) protective with 10%Al (FeCrAl)

Standard Oxidation Lifetime Model

Lifetime model developed by Quadakkers, Bennett, et al. for ODS FeCrAl alloys with 1-3mm cross-sections

Premise: Calculate time to breakaway (FeO_x formation) by knowing total **Al reservoir** available and rate of **Al consumption**

Model inputs:

- initial Al content (C_0)
- the **critical Al content where Al_2O_3 will no longer form**: (C_b)
- the thickness of the specimen (d) and density (ρ)
- Al consumption rate (e.g. kt^n), t is time,
 $n=0.5$ for parabolic, 1 for linear kinetics

$$(C_0 - C_b) / 100 \cdot d / 2 \cdot \rho = k \cdot t^n \cdot \frac{\text{(mole Al)}}{\text{mole O in Al}_2\text{O}_3}$$

How does this apply to a coating?

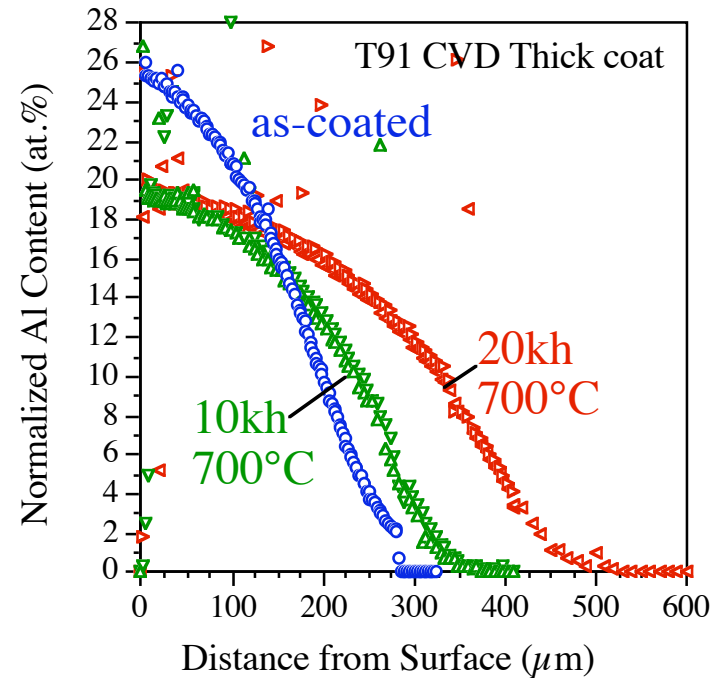
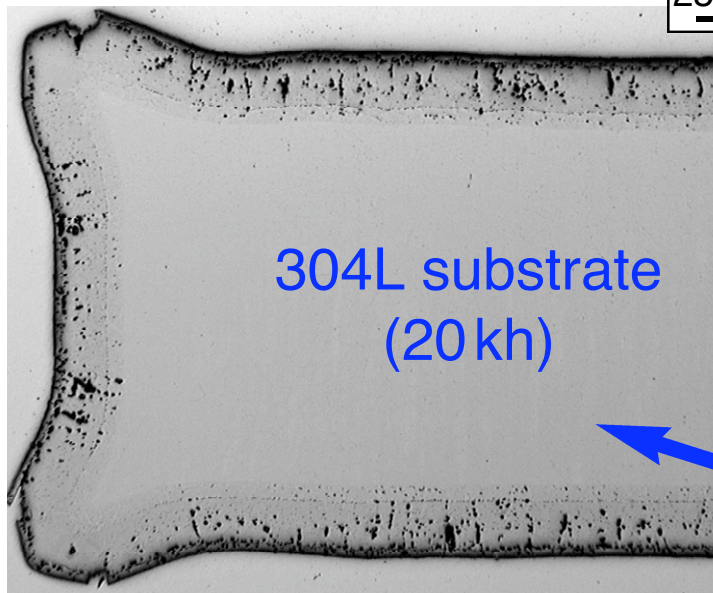
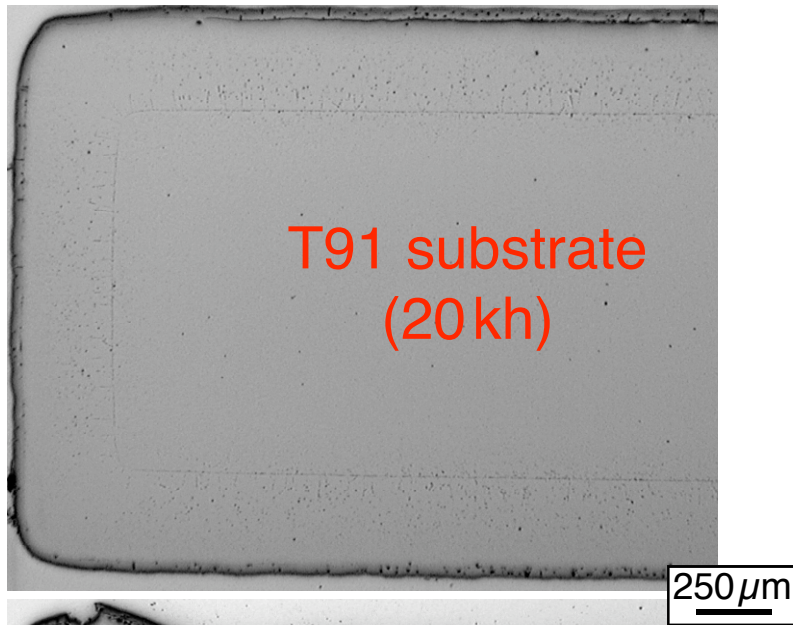
more complex Al “consumption”: **interdiffusion + oxidation**

C_0 becomes a function of the coating thickness

What is C_b for a coating?

700°C Performance of thick coating

Coatings stopped after 10 & 20kh in humid air



T91 substrate

Coating Al profiles on T91:

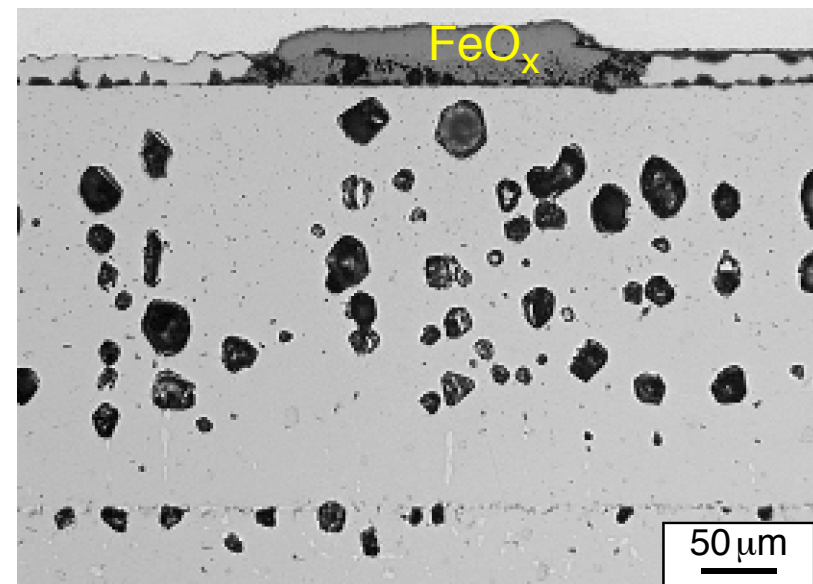
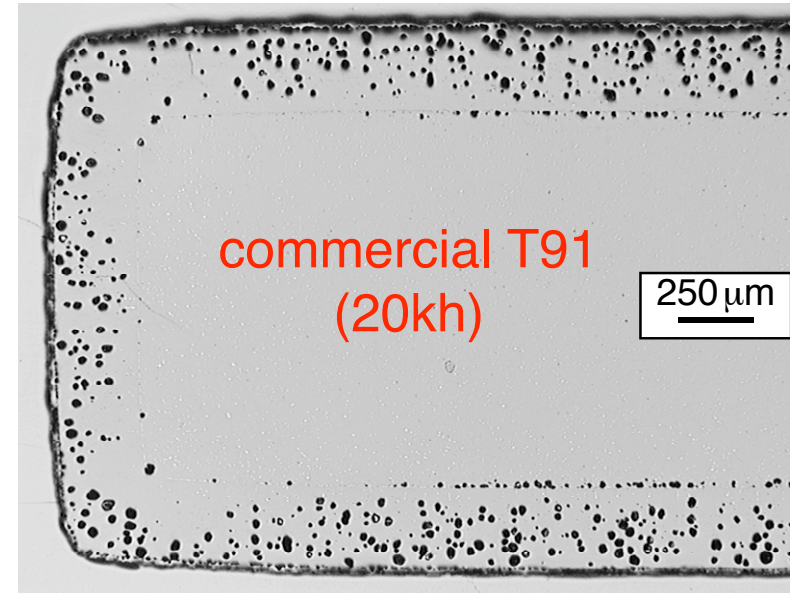
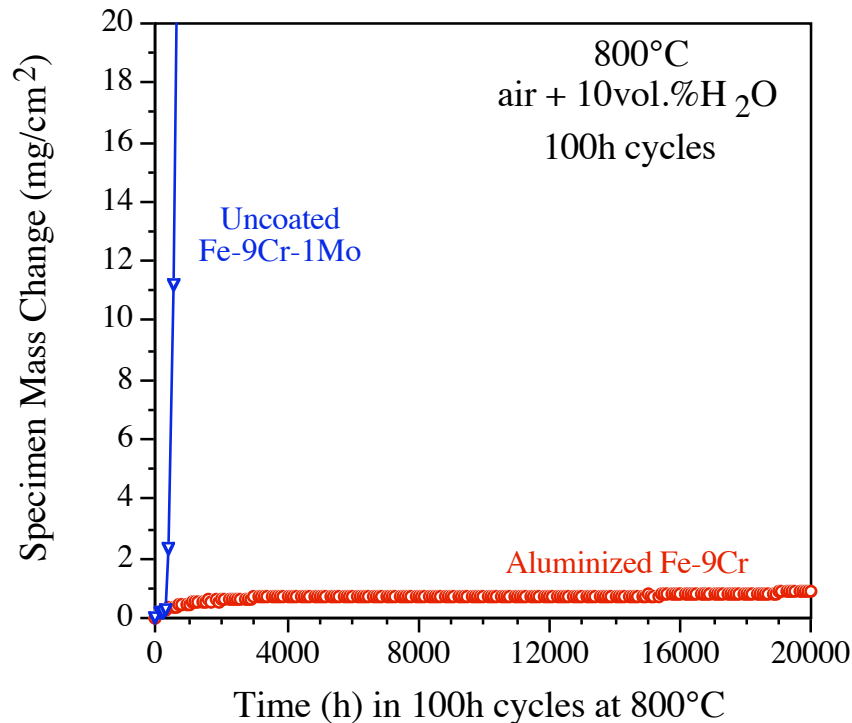
- all for thick coatings
- variations in starting thickness

Deformation difference:

- Δ CTE difference for 304L

20kh Coating Characterized

Thick coating on T91 in humid air at 800°C



20kh specimen:

No macroscopic deformation

Low mass gain

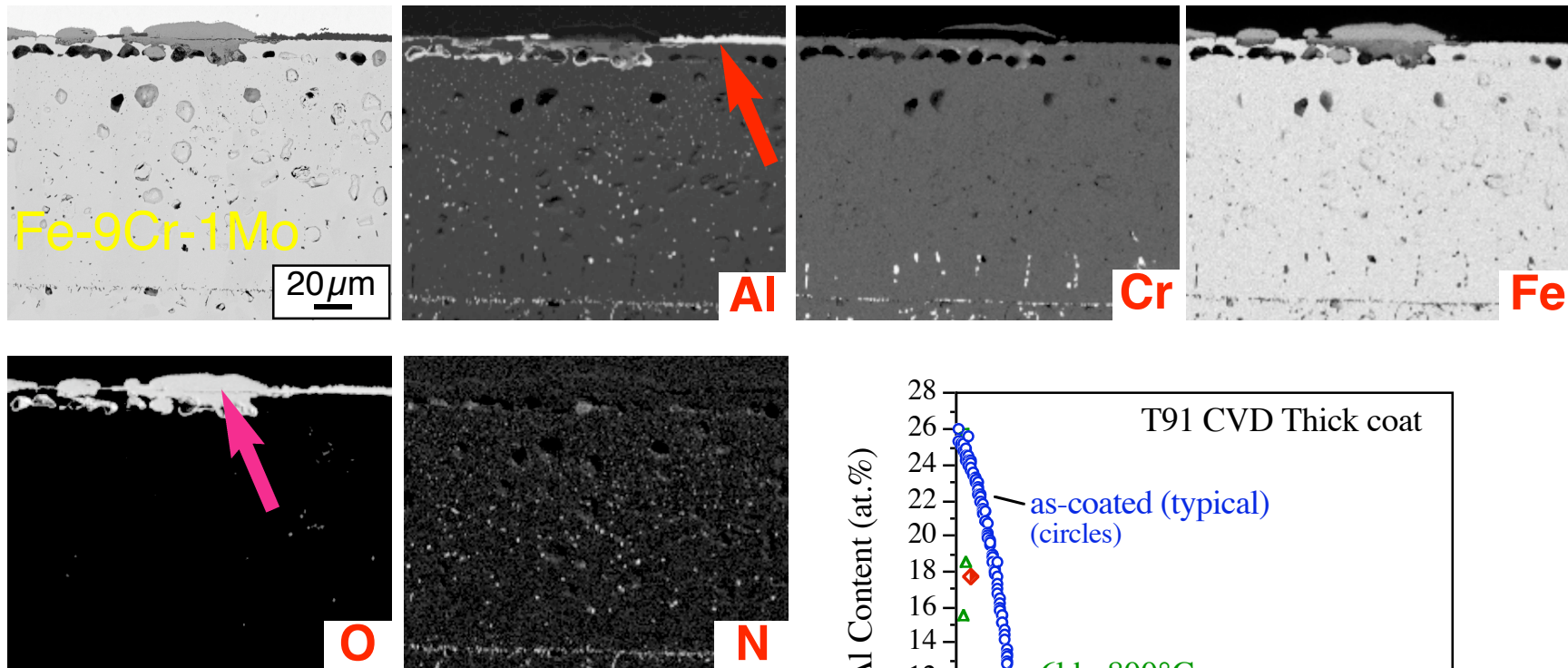
- but outer layer local breach

Porous coating layer

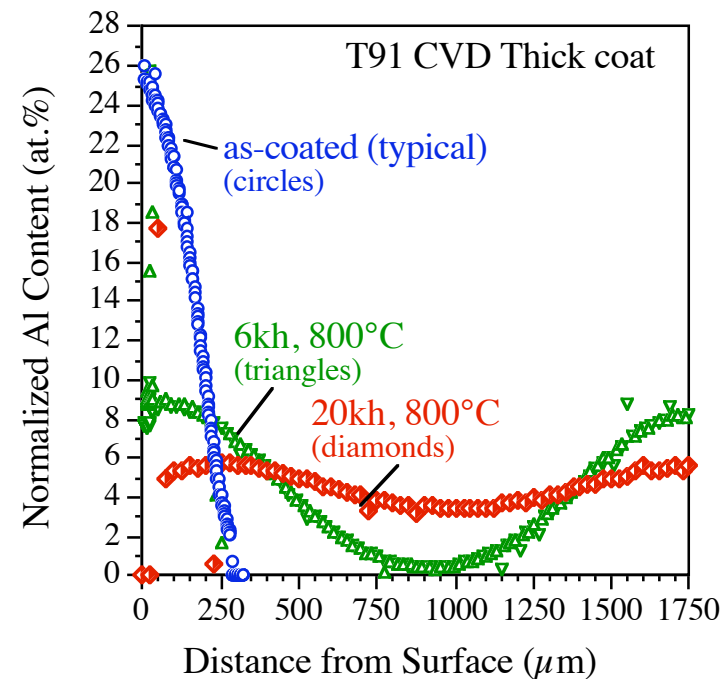
- Al loss due to scale spallation

20kh Coating Characterized

Thick coating on T91 in humid air at 800°C

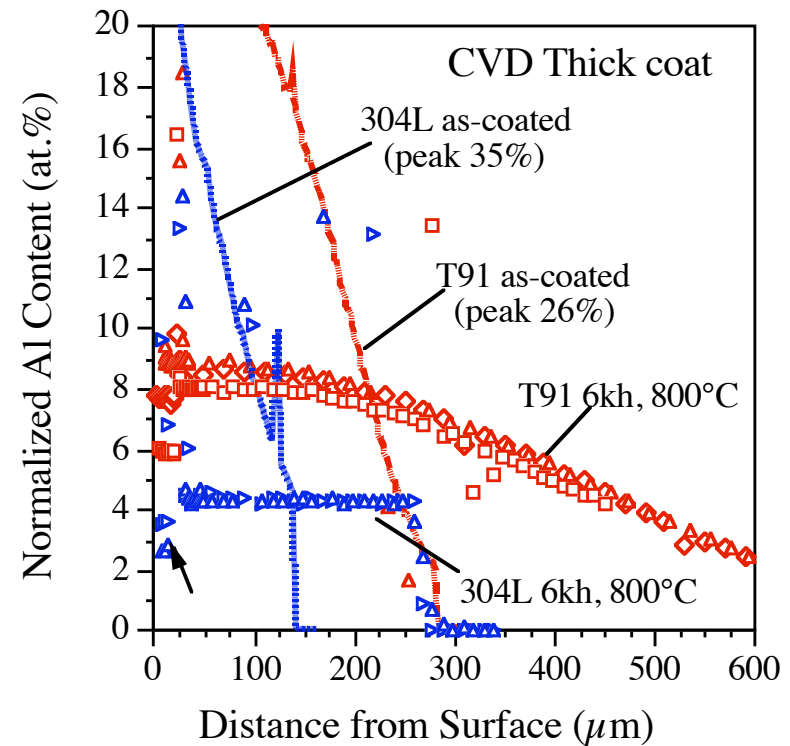
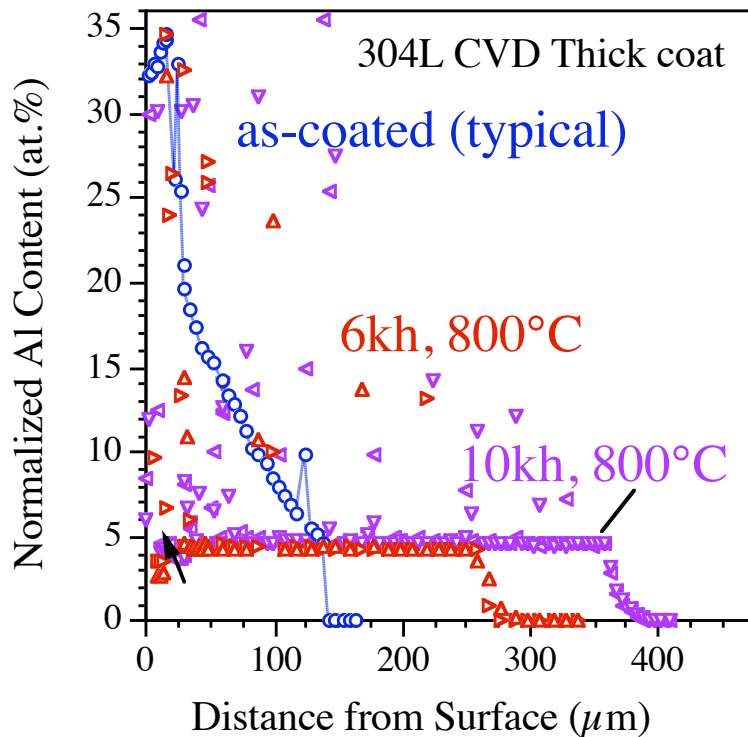


Electron probe analysis:
Typical Fe-rich oxide nodules
Local alumina scale pieces
AlN precipitates (0.2at.%N in alloy)



Future work: austenitic model

Four component system (Fe, Ni, Cr, Al) + two phase



304L: COSIM model missing diffusion terms

Three phase system (β -(Fe,Ni)Al+ ferrite + substrate)

Observations: Thinner starting coating than on T91

Slower Al diffusion: inhibited by phase transformation

$\sim 4.5\text{at}\%\text{Al}$ remained in inner layer (+ $\sim 18\%\text{Cr}$) - equilibrium?

Lifetime predictions at 700°C

Calculations for $\sim 250\mu\text{m}$ thick coatings on T91

Prediction Method	COSIM dependent	COSIM indep. FeAl	COSIM indep. FeCrAl	Heckel	Actual (diffusion test)
Surface (at.%)	19	16	17	19	18%Al
Thickness (μm)	310	438	428	356	320 μm

Lifetime Predictions:

(sulfidation)

assuming 20% 6.8 kh 5.0 kh 5.6 kh 7.5 kh ?

(wet air)

assuming 3.5% 639 kh 219 kh 248 kh 592 kh ??

assuming 8% 187 kh 57 66 104 kh ?

(conservative)

Sulfidation - insufficient life at 700°C, need to drop to $\sim 625^\circ\text{C}$

Wet air - high probability of thick coating making 100kh lifetime

Model details in Zhang et al., Mater. Corr. 58 (2007)

Lifetime predictions at 700°C

Calculations for $\sim 250\mu\text{m}$ thick coatings on T91

Prediction Method	COSIM dependent	COSIM indep. FeAl	COSIM indep. FeCrAl	Heckel	Actual (diffusion test)
Surface (at.%)	19	16	17	19	18%Al
Thickness (μm)	310	438	428	356	320 μm

Lifetime Predictions:

(sulfidation)

assuming 20% 6.8 kh 5.0 kh 5.6 kh 7.5 kh ?

(wet air)

assuming 3.5% 639 kh 219 kh 248 kh 592 kh ??

Sulfidation - insufficient life at 700°C, need to drop to $\sim 625^\circ\text{C}$

Wet air - high probability of thick coating making 100kh lifetime

- thick coatings will never fail on 1.7 mm substrate!